

1 **Geochemical, mineralogical and geostatistical modelling of an IOCG**
2 **tailings deposit (El Buitre, Chile): implications for environmental safety and**
3 **economic potential**

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23

24 **Abstract:**

25 In the mining industry deposits of tailings represent large volumes of the mining wastes,
26 reflecting the mineralogy and chemistry of the ore deposit type of origin. In this paper, we report
27 the results of a mineralogical and geochemical characterization study and geostatistical
28 modelling of an Iron Oxide Copper Gold deposit with neutral pH, in an arid climate. Twenty-eight
29 boreholes allowed recovery of 755 samples for analysis. Modelling by means of co-kriging
30 spatial interpolation allowed determination of the distribution of concentrations for elements of
31 interest in the deposit. Low water flow and near neutral-pH paste restrict the mobility of the
32 chemical elements, limiting the development of an oxidation front and inhibiting the appearance
33 of a secondary mineral enrichment zone and the precipitation of secondary efflorescent salts on
34 the tailings surface. Our observations determine that the composition of the gangue material,
35 and to a lesser extent the effect of the tailings gravitational deposition and the geometry of the
36 deposit, control the geochemical and mineralogical associations and distributions present in this
37 deposit of tailings. The observed low mobility along the tailings profile of some potential
38 pollutants (e.g., As, Mn, Cr and Ni) allows to anticipate a low groundwater pollution risk as long
39 as the current environmental conditions remain. Additionally, the depositional history of the
40 tailings had a great influence on the vertical and horizontal distribution of pyrite. However, the
41 grades of elements economic interest such as Fe, Cu and Co are uniformly distributed, which
42 would facilitate the application of tailings reprocessing technologies. In addition, it would allow
43 maximizing the number of necessary boreholes for its economic evaluation.

44 **Keywords:** Fe oxide Cu–Au deposits, Tailings, Conceptual model, Neutral pH, Arid climate.

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48 **1. Introduction**

49

50 The deposits of tailings could contain multiple elements considered to be of strategic interest,
51 which were not extracted in the past, either because they were not considered to be of
52 economic value, or due to the lack of appropriate technologies for their economic recovery
53 (Lottermoser, 2010; Pan et al., 2014). In addition, a recent study by Franks et al., (2021) have
54 estimated the presence of more than 8,100 tailings deposits globally and the generation of 10
55 billion m³ of fresh tailings every year. As a result, deposits of tailings have become a focus of
56 interest as possible alternative sources of raw materials to partially mitigate the shortages on
57 many critical raw elements imposed by the always growing demand of technological
58 applications and green energies (Parviainen et al., 2020). The economic potential of these types
59 of waste has been demonstrated in various tailings worldwide (Araya et al., 2021; Araya et al.,
60 2020; Cenicerós-Gómez et al., 2018; Dino et al., 2018; Falagán et al., 2017; Moran-Palacios et
61 al., 2019; Pan et al., 2014; Parbhakar-Fox et al., 2018; Parviainen et al., 2020). Complementary,
62 there are also numerous studies dealing with the environmental characterization of mine tailings
63 (Buch et al., 2021; Cleaver et al., 2021; Lam et al., 2020, 2021) and proposing different
64 reutilization alternatives, such as transforming them into ceramics and construction materials
65 (Kang et al., 2021; Krishna et al., 2021; Zhang et al., 2021).

66

67 The deposits of tailings exposed to the environment are susceptible to weathering. In
68 consideration of supergene, processes that affect these deposits, and based on present
69 knowledge the following aspects should be considered:

70

71 1) The presence of minerals such as calcite, dolomite and, to a lesser extent, some
72 aluminosilicates, can effectively neutralize the acidity generated and maintain circum-neutral pH
73 conditions (Blowes et al., 1998; Lindsay et al., 2009a);

74 2) The mobility of liberated elements by oxidation, leaching and neutralization reactions
75 depends on the climatic conditions, the geometry of the deposit and multiple biogeochemical
76 processes, including precipitation/dissolution, or redox;

77 3) Stratifications within the deposits of tailings are determined by: *i*) the mineralogy of the
78 source reservoir, *ii*) the grain size distribution and *iii*) ore mineral liberation resulting from the
79 mineral processing stage, and *iv*) the gravitational segregation processes during the settlement
80 of the tailings (Dold and Fontboté, 2001, 2002; Kovács et al., 2006; Heikkinen and Räisänen,
81 2008; Marescotti et al., 2010; Pan et al., 2014; Wang et al., 2017).

82
83 Several characterization studies have been carried out on tailings from sulfide-rich mineral
84 deposits, such as copper porphyry, skarn, volcanogenic massive sulfide, among others (Dold
85 and Fontboté, 2001; Lindsay et al., 2009a; Lindsay et al., 2009b; Hällström et al., 2018). Dold
86 and Fontboté (2001) determined that the deposit of tailings generated from copper porphyry
87 deposits have a low neutralization potential associated with the low carbonate content and the
88 relatively high content of minerals derived from supergene enrichment. These characteristics,
89 together with favorable climatic conditions and long exposure time of the tailings typically lead to
90 the development of acid mine drainage (Smuda et al., 2014). Skarn-type deposit materials are
91 considered high in carbonates, sulfides, and fluorite. As a result, their deposit of tailings
92 frequently exhibit high mobility of metals due to the pH decrease induced by iron-sulfides
93 oxidative dissolution (Blowes et al., 2003) and complexation with F ions keeping metals mobile
94 (Petrunic and Al, 2005). On the other hand, tailings deposits with low-sulfur and high-iron
95 (Henne et al., 2020), like the deposits generated after the mineral processing of Iron Oxides
96 Copper Gold (IOCG) or Iron Oxide-Apatite (IOA) type deposits have been barely studied (Dold
97 and Fontboté, 2002; Henne et al., 2019; Medina et al., 2019) and their weathering and evolution
98 in arid and semi-arid climates is poorly understood.

99

100 IOCG-type ore deposits are globally distributed, grouping different styles of mineralization.
101 Only eleven IOCG provinces “sensu stricto” have been recognized worldwide, including the
102 Central Andean Coastal Belt, spanning from Southern Peru to Northern Chile (Sillitoe, 2003). In
103 recent decades they have been considered profitable exploration targets due to the economic
104 recovery of copper, gold and other by-products such as uranium (Barton, 2014). These deposits
105 may also contain high concentrations of rare earth elements (REEs), silver, molybdenum,
106 nickel, cobalt, barium, fluorine, and phosphorus (Barton, 2014). Most of these elements are on
107 the European Commission’s list of critical raw materials due to their economic importance and
108 supply risk (European Commission, 2017).

109
110 In Chile there are more than 757 deposits of tailings, of which 173 are abandoned and 467
111 are inactive. More than 80 % are located in the north of the country where the climatic
112 conditions are arid and semi-arid and where the main IOCG deposits are located (e.g., Manto
113 Verde, Punta del Cobre, Cerro Negro, El Espino and Candelaria) (Sernageomin, 2020a). Due to
114 the difficulties and costs of drilling these deposits and the low economic potential they are
115 thought to represent, most tailings studies in Chile (and around the world) are usually limited to
116 near surface samples, depths of 10 m or less. However, there are compelling reasons that
117 justify obtaining a better knowledge of the deeper sections of these deposits, among others:

118
119 1) In almost all cases there is a complete absence of historical record of the mineralogical
120 and geochemical characteristics of the tailings deposited during the years and decades of
121 production, being impossible to anticipate the existence and extent of sedimentation layers with
122 different mineralogical and geochemical characteristics within the different depths of the tailings.

123 2) Old deposit of tailings (with many years and even decades after their closure) typically
124 generate alteration fronts (e.g., leaching, enrichment and/or depletion of certain elements) that
125 may evolve deep into the deposit of tailings.

126 3) Depending on the geometry of the tailings deposit, during its deposition the accumulation
127 of heavy minerals in the deeper layers can be favored by gravitational processes, being
128 relatively enriched in heavy metals with respect to the shallower layers. Therefore, a detailed
129 understanding of the deeper sections of the deposit of tailings is essential to understand their
130 geochemical and mineralogical evolution, to generate realistic environmental assessments and
131 to create acceptable economic evaluations for a future revalorization of these secondary
132 deposits.

133
134 The main objective of this study, based on detailed and systematic sampling of an IOCG
135 tailings deposit and its geochemical and mineralogical characterization, is to develop
136 geostatistical (using interpolation by co-kriging) and conceptual models allowing to decipher its
137 geochemical and mineralogical spatial-temporal evolution as well as the most relevant physical
138 and chemical parameters controlling this evolution. To this end, the presence of alteration fronts
139 and their possible vertical and horizontal extensions were evaluated. Subsequently, the different
140 factors promoting the alteration of sulfide minerals under arid climate conditions were assessed.
141 In addition, the influence of the deposit geometry and the possible sediments depositional
142 history has been determined. Finally, the environmental implications and possible economic
143 potential of the deposit based on critical raw elements such as Co, REEs, Cu and Fe were
144 considered. This study places special emphasis on the mineralogical and geochemical
145 associations characterizing the final deposit of tailings, together with other relevant physical and
146 chemical parameters (e.g., paste pH, hydraulic conductivity, and particle size distribution).

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148

149 2. Site description

150 2.1 Ore geology

151

152 The Punta del Cobre district includes the Candelaria deposit, and several medium and small
153 sized mines such as Alcaparrosa, Carola, Las Pintadas, Socavón and Santos, all hosted within
154 volcanic rocks of the Punta del Cobre Formation. These rocks of a pre-upper Valanginian age
155 (Marschik and Fontboté, 2001). Rocks of this formation are subdivided from bottom to top into
156 the Geraldo-Negro Member (> 300 m) and Algarrobo Member (up to > 800 m) comprised mainly
157 of altered sub-aquatic volcanic and volcanoclastic rocks. Mineralization occurs in a diversity of
158 styles including veins, stratiform orebodies, stockworks and hydrothermal breccias, hosted by
159 andesite volcanic, volcanoclastic and volcanic breccia rocks, locally termed an albitophire facies
160 of the Punta del Cobre Formation (Marschik and Fontboté, 2001). Five main alteration types are
161 identified and associated with mineralization as well as a regional contact metamorphism
162 caused by the Coastal Batholith in the western sector. An early episode of hydrothermal
163 alteration generated extensive albitization (albite–quartz–chlorite, \pm sericite \pm calcite) which was
164 locally superimposed by potassic alteration (K-feldspar–quartz–chlorite/biotite \pm sericite \pm calcite
165 \pm tourmaline). The rest of the alterations are associated with the batholith and are characterized
166 by the following mineral assemblage (from contact, west to east): Ca-amphibole \pm biotite \pm
167 sericite, biotite \pm chlorite \pm sericite \pm epidote and epidote chlorite \pm quartz \pm calcite (Marschik
168 and Fontboté, 2001). Main hypogene minerals in the Punta del Cobre District are chalcopyrite,
169 pyrite, magnetite and hematite (Oyarzun et al., 1999).

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171

172

173 *2.2 Description of the studied mine tailings*

174

175 El Buitre tailings deposit is located in the province of Copiapó, in the community of Tierra
176 Amarilla. The General Directorate of Water of Chile has reported the presence of an unconfined
177 aquifer consisting mainly of an alluvial deposit. The aquifer has been subdivided into 6 sectors
178 to facilitate management of its resources (Fig. S1). El Buitre tailings deposit is located in the
179 vicinity of sector IV, in the lower part of the sub-basin of the Middle Copiapó River, specifically in
180 Los Buitres and Los Diques sub-basins (Compañía Contractual Minera Candelaria, 2013). This
181 region is characterized by an arid climate with an annual average temperature of 18.2 °C and an
182 average annual accumulated rainfall of 22.5 mm/year for the period of 1998-2019. The months
183 of June, July and August present the highest rainfall during the year. The record value in 24
184 hours was 92.3 mm during the month of December 1997, with a return period of 200 years.
185 However, there is a considerable number of years with null precipitation, or that the totality of
186 the annual precipitation was concentrated in one or two days of the year. The averages of
187 maximum, and minimum absolute temperature values in the same period were -0.4 °C and 40
188 °C (Ministerio de Obras Públicas de Chile, 2020). Evaporation rates strongly exceed rainfall
189 (Sistema de información integral de Riego, 2020).

190

191 At present El Buitre tailings deposit is closed, having been in operation from 1997 to 2005
192 and it has a total authorized tonnage of 6.6 Mt. After its closure, it was used until 2010 as an
193 emergency dam, in case the active Las Cruces tailings dam, currently active and belonging to
194 the Pucobre company, presented any inconvenience. Tailings in this deposit are the result of an
195 alkaline flotation circuit (~ pH 10.5-11) designed for the recovery of copper ore concentrates.
196 The plant processed ore deposits from the Punta del Cobre, Mantos del Cobre and Granate
197 Mines, however there are no reliable records regarding which specific mines the ore rock came
198 from. The Punta del Cobre mine extracts copper sulfides and oxides, while the Granate and

199 Mantos del Cobre mines extract copper sulfides. The deposit of tailings was built by the
200 downstream method, and the tailings were deposited in the form of slurry. According to the
201 historical image of the deposit of tailings (Fig 1a), multiple outfall deposition (spigots) was the
202 deposition method used. At least 3 discharge points are observed along the dam wall in the
203 eastern and central sector of the deposit. In addition, the tailings were not treated with
204 amendments or any remediation technology. However, it has a 30 cm thick layer of coarser
205 tailings fraction (separated by hydrocyclones) on the surface to avoid its dispersion by wind.

206

207 **3. Samples and methods**

208 *3.1 Sampling and fieldworks*

209

210 All samples were collected in two campaigns during the summer to reduce any heterogeneity
211 caused by climatic conditions. Eight cores up to depths of 4 m were taken with a hand auger,
212 and twenty-four cores at different depths (1-49 m) were taken by sonic drilling. The depth in
213 each core depended on the geometry of the deposit of tailings. Undisturbed samples were
214 obtained by sonic drilling, and color, moisture, grading and degree of alteration and oxidation,
215 with emphasis on sulfide contents were recorded in situ (Fig. 1a). A total of 755 samples were
216 obtained from 1 m sample composites obtained from 755 m of core. These were sealed in
217 plastic bags and stored in ice-packed cooler boxes (8 °C), for later analysis. For the
218 geochemical analysis samples were oven-dried at 30 °C and homogenized in the Mining
219 Sustainability Laboratory of the Chile University.

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224 *3.2 Physicochemical parameters*

225
226 Paste pH was measured according to (Tremblay and Hogan, 2000) using a HI-5222 Hanna
227 pH meter, equipped with a HI-1053B electrode. The pH meter was calibrated every 30 samples
228 with certified standard reference solutions (HI7004L/C and HI7007L/C). In addition, duplicates
229 were analyzed every 10 samples. In addition, particle size distribution of a representative
230 sample composite was performed by Cisa Ro-Tap sieve shaker at ASMIN industrial laboratory,
231 Santiago, Chile. The hydraulic conductivity was measured by a constant-head method for the
232 laminar flow of water through granular soils following the ASTM D2434 procedure. The visual
233 inspection of the drill cores allowed to qualitatively differentiate between the dry and wet tailings
234 samples. This information was used to define a “moisture horizon” that could be defined by the
235 depth at which wet tailings samples are unequivocally identified in each drill core.

236
237 *3.3 Geochemical methods*

238
239 *3.3.1 Total digestion*

240 The 755 samples collected from 1 m sample composites were completely digested using a 4
241 acids digestion procedure ($\text{HNO}_3 + \text{HF} + \text{HClO}_4 + \text{HCl}$). A total of 59 elements were measured by
242 ICP/MS (Perkin Elmer Elan, 9000) with detection limits as indicated in Supplementary material,
243 Table S1. One blank was run for every 40 samples, and an in-house control was run every 20
244 samples. For every 15 samples, a digestion duplicate was analyzed. In addition, digested
245 standards and instrument recalibration were run every 80 samples. All the sample digestions
246 and analyses were performed by Actlabs, an accredited laboratory at Coquimbo, Chile.

247

248 3.3.2 *Sequential extraction*

249

250 The sequential extraction was performed by AGQ Laboratories, Santiago, Chile, following the
251 procedure defined by Dold and Fontboté (2001) with the objective to study element speciation in
252 the deposit of tailings. Three boreholes (B-04, B-13 and AD-02) located in the eastern and
253 central sectors were selected with the objective of studying element speciation to the deepest
254 zones of the deposit. A total of 17 composite samples every 2 m were selected from the most
255 superficial, intermediate, and deep zones of the deposit. The depth ranges were selected,
256 taking as criteria mineralogical changes detected from the detailed analysis of the boreholes
257 core and variations in the geochemical data with depth. The first four extractions allowed to
258 separate the mobilized elements and secondary minerals. The last three stages allowed to
259 separate organic material, secondary and primary sulfides, and silicates. For more details on
260 the physical and chemical treatment of the samples and the main dissolved minerals in each
261 step, see Table S2. The solutions were analyzed by Inductively Coupled Plasma Atomic
262 Emission Spectroscopy (ICP-OES) with detection limits indicated in Table S3. Saturation index
263 ($SI = \log IAP - \log K$; IAP = ion activity product) of gypsum with respect to water soluble fraction
264 element concentrations were calculated using PHREEQC version 3.6.2-15100 and the paste pH
265 value was used as a reference.

266

267 3.4 *Mineral characterization methods*

268

269 A total of 340 composite samples every 1 m were analyzed by X-ray diffraction (XRD).
270 Samples located in the closest and most distal zone of the dam wall, and in the central zone of
271 the deposit of tailings were analyzed. In addition, samples belonging to different depth ranges
272 were selected, taking as criteria mineralogical changes detected from the detailed analysis of

273 the boreholes core and variations in the geochemical data with depth. These selection criteria
274 were also used for all other mineralogical analyzes. Homogenization and micronisation of the
275 tailing's particles were carried out to obtain grain sizes of less than 10 microns. The
276 mineralogical composition was determined using a Bruker® D8 Endeavor X-ray diffractometer
277 with a tube with cobalt radiation. Diffractometer settings were: 35 kV, 40 mA and a scan range
278 of 3–70° 2 θ , 0.02° 2 θ step size, and 5 s counting time per step. The obtained diffractograms
279 were analyzed using the software DIFFRAC. EVA® version 4.2 and DIFFRAC.TOPAS® version
280 5.0. The Powder Diffraction File™ (PDF-2 2001) database was used for phase identification.
281 Spectra standard of corundum was used for semi-quantification of mineral phases. Besides,
282 QA/QC was performed by analysis of duplicates and X-ray fluorescence. In addition, clay
283 minerals from 202 composite samples every 1 m were characterized following USGS protocol
284 (USGS, 2000). The < 2 μ m fraction was separated by centrifugation and the samples were
285 analyzed by XRD before and after heat treatment. The semi-quantification of clays was carried
286 out with spectra standards of montmorillonite, kaolinite and illite. XRD analyzes were performed
287 by Geomaq Limitada, an accredited laboratory at Antofagasta, Chile.

288
289 A total of 100 composite samples every 1 m were prepared in polished thin sections and
290 analyzed by optical microscopy Olympus model BX51, using reflected-light mode to identify
291 opaque minerals in the tailings assemblage. The mineral composition was determined by the
292 count point method at 350 equally spaced points across the thin sections. In addition,
293 photomicrographs were obtained to complement the observations. This analysis was made in
294 the Minerals Laboratory of the Mining Department of the Chile University. In addition, semi-
295 quantitative element composition was studied by Quantitative Evaluation of Materials by
296 Scanning Electron Microscopy (QEMSCAN) combining information generated by the retro
297 dispersed electrons (for imaging) and X-ray dispersive energy spectra (EDS) for chemical
298 composition. A total of 150 composite samples every 1 m were analyzed. The Bulk Mineral

299 Analysis (BMA) measurement method was used to provide statistically abundant data for
300 mineral identification, speciation, distribution, and quantification. This analysis was performed by
301 SGS minerals, Chile.

302

303 *3.5 Statistical methods and geostatistical modelling*

304

305 The database of total digestion and sequential extraction was filtered and adjusted according
306 to the following criteria. 1) All the values below the detection limit (LOD) were changed
307 according to the values resulting from dividing the specific detection limit value by the square
308 root of 2 (Hites, 2019). 2) Variables with more than 50 % of data under the LOD was not
309 suitable to be included in any statistical analysis and was eliminated. 3) The normality of the
310 data was evaluated by the Kolmogorov-Smirnov tests. Due to the non-normal distribution of the
311 dataset, the transformation to normal logarithm was performed. 4) Alpha was set at 0.05 for
312 interactions. The mean value of chemical data (as the dependent variable) concerning different
313 depths was analyzed using one-way factorial ANOVA and Welch t-test. Tukey's and Dunnett
314 T3's multiple comparisons procedures by homogeneous and non-homogeneous variables were
315 performed.

316

317 A Principal component analysis (PCA) was carried out by means of a Pearson's correlation
318 matrix (0.05 significance level, 2-tailed test) on all validated variables in order to define the main
319 element associations and the reflected mineralogical variations, for the interpretation of
320 processes such as weathering or to characterize variations of origin. This analysis is also used
321 to interpret these associations in terms of geochemical processes by comparing their patterns
322 with other analytical determinations (sequential extraction, mineralogical characterization, and
323 geostatistical model). PCA discovers linear combinations of the variables based on measures of
324 association and, it allows the reduction of dimensions for a complex multi-element dataset. The

325 results are graphed in a biplot. It is a plot of the PC loadings for each variable with the loadings
326 of each sample along two sets of PC axes (Abraitis et al., 2004). Evaluation of the chemical
327 data, PCA analysis and one-way ANOVA was performed using ioGAS 7.3 and IBM SPSS
328 Statistics 25 software, respectively.

329

330 The estimation of the grades in the deposit of tailings was carried out using interpolation by
331 cokriging. This method allows modelling the grades in the deposit based on drill core depths
332 and considering together all the grades of the study and the spatial continuity through
333 variographic analysis. Due to the lack of information of the previous underlying topography the
334 tailings, the tailings bottom surface was modeled by kriging the length of the drill cores at the
335 studied domain. The estimation of the bottom surface was performed as follows: 1. the bottom
336 surface was defined to estimate the total volume of the restricted modeled area (Kokkola, 1986);
337 2. the bottom surface of the whole the deposit of tailings was restricted by the dam structure, to
338 estimate the total volume of the deposit (Parviainen, 2009). The total resources of Co, Cu, light
339 rare earth elements (LREE), heavy rare earth elements (HREE), Fe, S, Ca, P and Al were
340 estimated using a 3D block model and a discretization cell of 3 m x 3 m x 2m (Deutsch and
341 Journel, 1997). This model was arranged according to a sampling grid with horizontal distances
342 of five meters and depth intervals of one meter (block size of 5 m x 5 m x 1 m) resulting in a
343 total of 1,800,000 blocks. The validation of the models was carried out by cross validation with
344 the chemical data. The modelling workflow was performed by the method proposed in
345 (Parviainen et al., 2020). This methodology takes into consideration the difficulties in defining
346 the topography of the abandoned tailings and the anisotropy generated by the greater length of
347 the horizontal direction than the vertical direction. Multivariate modelling was performed using
348 the ANDES software (Soto et al., 2017).

349

350 4. Results and Discussion

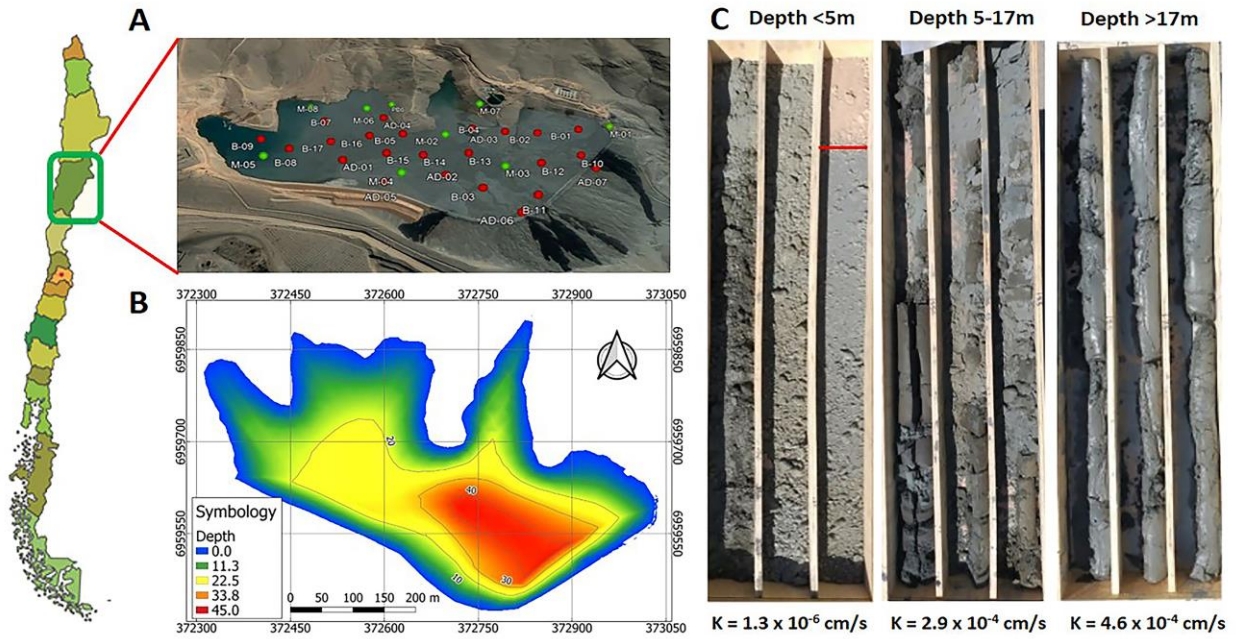
351 4.1 Tailings morphology and physical and hydraulic main characteristics

352

353 Modelling the tailings' depth and current surface topography is very important to properly
354 estimate the tailings precise volume for future revalorization and reprocessing studies. It is also
355 important to control the physical stability of a future operation because it will control the
356 formation of a ditch network eroding the tailings surface (Parviainen et al., 2020). The thickness
357 of the deposit of tailings was determined from the drill holes, taking into consideration that the
358 surface topography was quite regular, and the deepest areas were delimited by coarse sand
359 and gravel and by a waterproof membrane of high-density polyethylene in the area the dam
360 wall. The deposit's thickness varies from one to 49 m. The deepest parts of the tailings are in
361 the eastern and central parts of the deposit of tailings (Fig. 1b), following the original shape of
362 the valley basin where the tailings were deposited. In the western area, the depth tends to
363 decrease, ranging from 1 to 25 m (Fig. 1b). From a sedimentary perspective, visual field
364 observations showed that the tailings deposit is comprised by alternating fine sandy to clayey-
365 silty layers. Particle size analyses of a composite sample from one drill hole showed that the
366 cumulative passing curve is ranging from 1 to 300 μm , and d_{80} is 40 μm (Fig. S2 a and b). This
367 distribution of particle size is in agreement with the milling stages required for the flotation
368 process in tailings with similar characteristics (Medina et al., 2019). Please notice that, although
369 this grain size distribution curve can be useful for a general understanding of the tailings
370 deposit, it does not provide detailed information about the grain size distribution within the
371 specific horizontal layers observed along the tailings depth profile. The proportion of sand and
372 clay changes with borehole depths and with their locations respect the dam wall. The first five
373 meters of depth are typically more sandy, whereas the intermediate zone (5-17 m) presents
374 intercalations of sand-clay-sized material, and the deepest zone (> 17 m) is mainly clayey (Fig.

375 1c). This particle size distribution has a direct impact on the hydraulic conductivity of the deposit
376 of tailings (Fig. 1c), resulting on low hydraulic conductivity $K = 1.3 \times 10^{-6}$ cm/s for the deepest
377 layer and higher hydraulic conductivities (in the order of 10^{-4} cm/s) for the more surficial layers
378 (0-17 m). To illustrate the distribution of the moisture horizon within the deposit of tailings, an
379 iso-piezometric map was made. It shows how this horizon is deeper towards the central sector
380 of the deposit of tailings, in the area more distal to the streams located northeast and northwest
381 of the deposit of tailings (Fig. S3). On the other hand, the moisture content within the deposit of
382 tailings appears not to be influenced by the presence of groundwater. According to groundwater
383 level monitoring and quality control wells located in El Buitre sector (Fig. S1), the confined
384 aquifer is found at a depth greater than 125 m (the maximum tailings depth is 49 m). Besides,
385 the Alcaparrosa well located in the vicinity of El Buitre sector, has diminished by 40 m in the
386 period 1998-2008, as a result of agricultural activities in the area. Additionally, the recharge of
387 the aquifer by natural processes is complex, since the potential evaporation considerably
388 exceeds the precipitation, causing the net recharge to tend to zero. Therefore, groundwater
389 recharge occurs during wet years. According to the record of the last 40 years of the Copiapó
390 Weather Station (1971-2011), it is estimated that the recharge is between 5 to 12 mm/year
391 (considering a recharge rate of 10 %) (Compañía Contractual Minera Candelaria, 2013).

392



393

394 Fig. 1. a) Historical satellite image (2004) showing tailings discharge points and the location of the
 395 boreholes (red spots = sonic drilling and green spots = hand auger); b) depth spatial variation within the
 396 deposit of tailings; c) representative tailings core sample boxes of different depth ranges. The first box on
 397 the right shows a light brown layer that corresponds to coarse tailings fractions used to avoid wind borne
 398 dispersion.

399

400 4.2 Mineralogical characterization

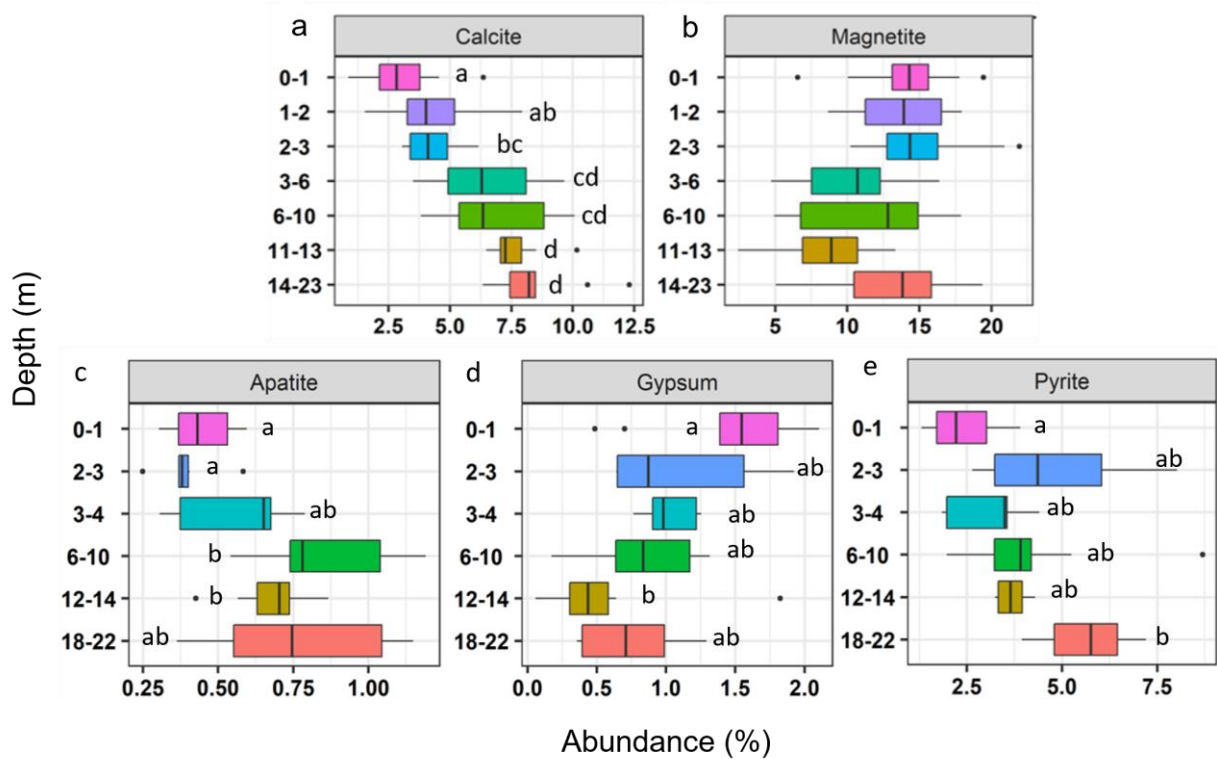
401

402 The mineralogical composition of the tailings is quite homogeneous, most samples are similar
 403 in mineral composition and abundance. XRD shows that quartz, chlorite, alkali-feldspar
 404 (microcline and orthoclase), plagioclase and magnetite are the main mineral constituents,
 405 whereas other minerals like hematite, calcite, pyrite, and gypsum-anhydrite appears as minor
 406 mineral constituents of the deposit of tailings (Fig. 2 and Fig. S4). Illite, chlorite, and kaolinite
 407 were the typical minerals in the clay fraction <2 μm. Additionally, QEMSCAN analyses (more
 408 reliable than XRD analyses for minerals with wt % lower than 1 %) show that muscovite, biotite,
 409 tourmaline, rutile, apatite, amphibole, sphene, pyroxenes, ilmenite, epidote, dolomite/ankerite

410 and garnet could be considered as trace constituents (Table S4). Also, the QEMSCAN study
411 revealed that chalcopyrite is the main Cu-bearing mineral (ranging from 0.163 wt % to 0.02 wt
412 %), whereas other minerals like chalcocite/digenite, enargite/tennantite and malachite/azurite
413 can be considered negligible, with an abundance lower than 0,0001 wt % (Fig. S5).

414

415 The relative abundance of magnetite (Fig. 2b), orthoclase and microcline (Fig. S4 c and h) do
416 not significantly change with depth. Therefore, they are homogeneously distributed in the
417 deposit of tailings and alkaline-feldspar probably does not act as neutralizing agent. In contrast,
418 pyrite and calcite (Fig. 2a and 2e) progressively increase their abundance with depth, moving
419 from values around 1.5 wt % and 2.5 wt % on the tailings surface to values around 3.5 wt % and
420 7.5 wt % in the deepest layer, respectively. This trend might be the result of pyrite oxidative
421 leaching and acidity generation at the most surficial part of the tailings (where scarce rainwater
422 reaches) followed by calcite dissolution and leaching solution neutralization. In accordance with
423 that, the tailings showed a mean paste pH of 7.9, with values consistently ranging from 7.78 to
424 7.95 (Fig. S2c). On the other hand, the gypsum content remains practically constant with depth,
425 however, it decreases significantly in the depth range of 12-14 m. (Fig. 2d). This gypsum
426 distribution is in accordance with aforementioned pyrite oxidation and leaching process and
427 neutralization of acid solutions by calcite. Apatite shows relative abundances significantly
428 higher for the deeper samples (in the range of depths of 6-10 m and 12-14 m) than the
429 shallower samples (Fig. 2c). This distribution could be attributed to the gravitational deposition
430 of apatite particles in the deposit of tailings.

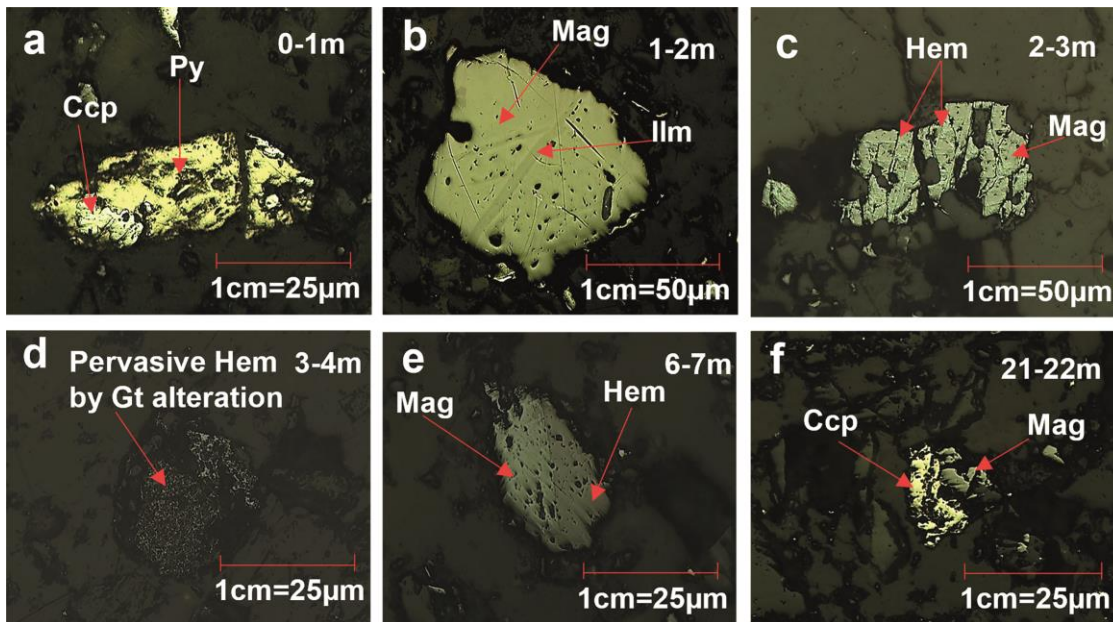


431
 432 Fig. 2. Mineralogical characterization of major minerals (calcite and magnetite) by XRD (340 samples)
 433 and minority minerals (pyrite, gypsum, and apatite) by QEMSCAN (150 samples) in samples at different
 434 depths. The lowercase letters on the graph represent a one-way ANOVA analysis. Different letters
 435 indicate that there is a significant difference between the relative abundance of the mineral at a certain
 436 depth. On the contrary, the same letters indicate that there are no significant differences. Minerals without
 437 letters indicate that the relative abundance of the mineral does not present significant differences in any
 438 of the analyzed depths.

439
 440 Complementary to the XRD study, the examination of polished sections showed that the
 441 sulfide assemblage is dominated by pyrite (ranging from 1.2 to 8.7 wt %), which is little fractured
 442 and disturbed in the oxidation zone, not showing Fe (III) oxyhydroxides rims or coatings (Fig.
 443 3a). Low alteration of the sulfide assemblage in neutral tailings has been reported by Dold and
 444 Fontboté, 2002; Lindsay et al., 2009b. Chalcopyrite and sphalerite are present as trace minerals
 445 and do not show replacements. Chalcopyrite is associated with magnetite, sphalerite, and pyrite

446 (Fig. 3a and f). Supergene Cu-sulfide were not identified by petrographic microscopy and
 447 QEMSCAN analysis indicated that these minerals are below 0.0001 % in abundance. The oxide
 448 assemblage is dominated by magnetite and hematite, and trace amounts of specularite,
 449 goethite, and ilmenite. Hematite in the form of specularite has been described by Marschik and
 450 Fontboté (2001). Magnetite is partially replaced by hematite (Fig. 3c) and goethite at the edges
 451 and fractures, and hematite is partly replaced by goethite. In addition, magnetite is commonly
 452 associated with ilmenite, hematite, and chalcopyrite (Fig. 3b, e, and f). The replacement of
 453 magnetite and hematite by goethite at the edges can occur for oxidation reactions in the deposit
 454 of tailings. On the contrary, the pervasive magnetite alteration to hematite and magnetite and
 455 hematite alteration to goethite (Fig. 3c and d) can be attributed to hydrothermal alteration
 456 processes within the deposit of origin.

457



458

459 Fig. 3. Mineralogical characterization by petrographic microscope in samples with different depths. A)
 460 Pyrite-Chalcopyrite (Py-CCp) association, b) Magnetite (Mag) with ilmenite (Ilm) bands, c) Mag partially
 461 replaced by Hematite (Hem) at the edges and fractures, d) Pervasive Hem by Goethite (Gt) alteration, e)
 462 Mag-Hem association and f) CCp-Mag association.

4.3 Geochemical and Geostatistical characterization

The original database of 59 elements (Table S5) was reduced to the 31 elements presented in

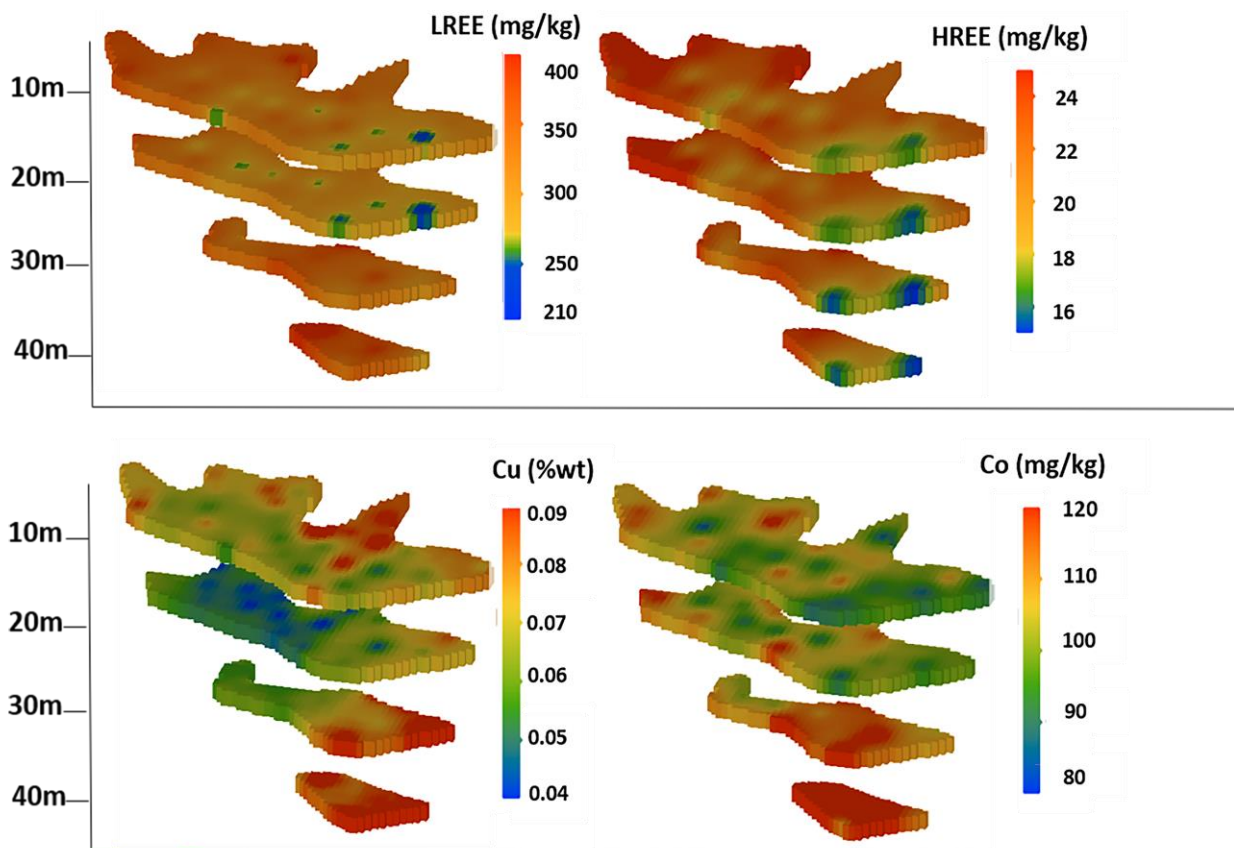
Table 1. In addition to the previous filters carried out before the statistical analysis, some immobile elements with very low concentrations were eliminated (Ba, Cd, Cr, Hg, Hf, Nb, Rb, Sb, Se, Sr, Ta, Te, Tl, Zr). REEs was divided into LREE and HREE to improve the visualization of the data distribution patterns and because of the very different commercial value of both groups. Table 1 summarizes the geochemical composition and the univariate statistics of the tailings. As expected from its mineralogical composition, the bulk chemistry of the tailings is mainly comprised of Fe, Si, Al, Ca, S and P, which is in accordance with the geochemical analyses of similar IOCG deposits in this region (Medina et al., 2019; Sernageomin, 2020b).

Taking as reference grades reported for ore deposits and deposits of tailings worldwide, it is possible to indicate a significant content of some value elements e.g., Co (103 ± 32 mg/kg), Cu (795 ± 514 mg/kg), and REEs (337 ± 118 mg/kg). In addition, it is necessary to take into consideration that due to the energy and infrastructure savings associated with the reprocessing of tailings, it is possible to exploit lower elemental ore grades (Parviainen et al., 2020). Mud and Jowitt (2018) reported Cu grades from 0.18 to 1.5 wt % in ore deposits around the World. While economic recovery of Cu has been reported in Minera Valle Central (Rancagua, Chile), a company that reprocesses fresh tailings from El Teniente mine and old mine tailings from the Cauquenes tailings deposit with Cu grades from 0.1 wt % - 0.25 wt % (Henderson, 2018). Regarding Co grades of ore deposits around the World, Mudd et al., (2013) reported values ranging from 0.019 to 0.24 wt %. While in the case of REEs, deposits ion adsorption clays contain REOs from 0.02-1 % (Su, 2009). On the other hand, some pollutants elements e.g., As (38 ± 12 mg/kg) and Pb (12 ± 6 mg/kg) are present in considerable concentrations in the solid fraction and their solubilization could represent an environmental risk.

489

490 A geostatistical model (using a 5 m x 5 m x 1 m block model, a total of 1,800,000 blocks) was
491 made to facilitate the interpretation and visualization (3D) of the bulk chemistry spatial variations
492 within the tailings (Fig. 4 and Fig. S6). In addition, it allowed to evaluate the tailings as a
493 “secondary ore deposit”, by modeling some critical elements, e.g., Cu, LREE, HREE and Co
494 (Fig. 4). Variographic analysis and validations of the model can be found in Fig. S7 and Fig. S8.

495



496

497 Fig. 4. Spatial distribution of LREE, HREE, Cu and Co grades in the deposit of tailings according to the
498 block model. All 3D views were separated into sections at every 10 m depths to facilitate data
499 visualization.

500

501 Based on the block model, some general trends for the mineral forming elements can be
502 observed. The Fe concentrations distinctly increases from West to East and from top to bottom

503 (Fig. S6). On the other hand, Al and Ca concentrations tend to decrease from West to East in all
504 depths, clearly displaying the lowest concentrations at the dam wall proximity (Fig. S6). Sulfur,
505 and phosphate to a lower extent, show a tendency to increase with depth (Fig. S6). Regarding
506 the modelled grades for some selected critical elements (Fig. 4), a modelled mean grade for
507 LREE of 324 mg/kg (ranging from 209.1 to 524.2 mg/kg) was obtained. Some enriched layers (>
508 350 mg/kg) can be observed in the central and western sectors at all depths. However, the
509 tailing's LREE concentration can be considered relatively homogeneous. On the other hand,
510 HREE modelled mean grade is 21.5 mg/kg, ranging from 14.6 to 29.1 mg/kg, showing more
511 significant grades increase on an east-western direction in all depths. Modelled Cu mean grade
512 is 0.069 wt %, ranging from 0.037 to 0.157 wt %. The highest concentrations are found in the
513 eastern sector at 30 m and at the bottom of the tailings (depth > 40 m, Fig. 4). Finally, the
514 modelled Co mean grade is 105 mg/kg (Fig. 4), ranging from 79 mg/kg to 171 mg/kg, and
515 clearly showing higher grades in the deepest levels (> 30 m).

516

517 It is important to take into account that the present geostatistical model is based on chemical
518 data obtained from a high density sampling campaign (32 boreholes) that were drilled to the
519 bottom of the deposit. This allows a reliable estimation of the grades and geometry of the
520 deposit. Therefore, the results of the present study can be considered an excellent exploratory
521 model to evaluate the economic potential of this waste. Based on the mean grades of the
522 elements of interest and the lowest reported cut-off grades for tailings and deposits worldwide, a
523 rough estimation of the total remaining resources for Cu, Co and REEs (Table 2) using the
524 volume of the whole the deposit of tailings was made. It was considered a tonnage of 6.6 Mt
525 (Sernageomin, 2021a). Additionally, Fe was included, because currently the Compañía Minera
526 del Pacífico (Copiapó, Chile) produces pellet feed with tailings from the copper concentrator
527 plant of the Compañía Contractual Minera Candelaria, with average Fe grades of 11.4 %
528 (Compañía Minera del Pacífico, 2020). According to this estimate, the modeled mean

529 concentrations of Fe and REEs present 100 % of the blocks above the cut-off grade (11.4 wt %
530 and 0.02 wt %) and they represent 1.023 Mt and 0.00132 Mt of the reserve (Fig.S9). The
531 modeled mean concentrations of Cu present an estimated 2.1 % of the blocks are profitable,
532 with grades above the cut-off grade (0.10 wt %). While the average modeled concentrations of
533 Co show 100 % of the blocks below the cut-off grade (0.19 wt %) (Fig S9). On the other hand,
534 economic potential was calculated as the fraction of each value element in the tailings multiplied
535 by the mass of the deposit and the metal price (Table 2). The price of REEs was 4,651,315
536 US/ton and it corresponds to the average of Eu_2O_3 , Gd, Tb, Dy, Er, Y, La, Ce, Pr, Nd, Sm.

537

538 The results indicate a high economic potential for iron and REEs. In contrast, Cu and Co
539 represent a low potential, because the modeled average grades are below the reference cut-off
540 grades (Table 2). Parviainen et al, (2020) suggest considering tailings deposits as polymetallic
541 deposits to increase their economic potential. However, the revaluation of tailings depends on
542 other factors like metal price, element-bearing mineralogy for elements of interest, metallurgical
543 recovery technologies, dimensions of the deposit of tailings (allowing scale economy), OPEX
544 and CAPEX costs (Araya et al., 2020, Parviainen et al., 2020; Araya et al., 2021), among many
545 other considerations. El Buitre tailings deposit can be considered small (6.6 Mt), which could
546 limit its profitability. Nevertheless, adjacent to its location, there are other tailings deposits
547 belonging to the Pucobre company, with similar characteristics, that could be reprocessed
548 together. Currently, the Compañía Minera del Pacífico extracts 3.5 Mton/year of Fe by magnetic
549 separation from the deposit of tailings belonging to the Compañía Contractual Minera
550 Candelaria. It has the potential to reprocess deposits of tailings located nearby, including El
551 Buitre tailings deposit. Additionally, during the cleaning process and concentration of magnetic
552 Fe could concentrate pyrite, chalcopyrite and apatite, therefore the extraction of Cu, Co and
553 REEs as secondary products could be favored. The Compañía Minera Valle Central extracts Cu
554 from tailings by froth flotation. Meanwhile, the Co associated with pyrite has been extracted by

555 bioxidation-flotation and bioleaching of sulfidic tailings of iron mines (Ahmadi et al., 2015;
556 Parbhakar-Fox et al., 2018). On the other hand, the possible association of REEs at different
557 mineral phases e.g., phyllosilicates and apatites, could hinder their extraction. Araya et al.,
558 (2020) conducted a techno-economic feasibility study of industrial-scale REEs recovery from
559 copper industry tailings, proposing chloride-based hydrometallurgical extraction technology
560 processes as a potential alternative to traditional capital intensive hydrometallurgical processes
561 based on high temperature and pressure (Onyedika et al., 2012). The study was carried out in
562 Chilean tailings deposits with characteristics similar to El Buitre. They determined a net present
563 value (NPV) to 20 years (income that an investment will generate in the future) of 672,987 USD,
564 exceeding the initial investment cost taken as a reference (342,514,448 US\$). In addition, the
565 internal rate of return (IRR) that refers to the discount rate at which the NPV of future cash flows
566 is equal to the initial investment, was 10.03 %. It was almost the same as the discount rate
567 chosen for the project (10 %), indicating that the project was not highly profitable. Nevertheless,
568 these calculations considered the average price of REEs of 22 US/kg for the year 2018,
569 including cerium, lanthanum, samarium, gadolinium, praseodymium, dysprosium, and yttrium
570 oxides. Additionally, they projected for the year 2022 an increase of 23 US/kg. However,
571 currently the average price of REEs in element form is 4651.31 US/kg, including Eu, Tb and Nd.
572 Therefore, the extraction of REEs can currently be considered viable taking as a reference the
573 COPEX, OPEX, production capacity and discount rate proposed by Araya et al., 2020.

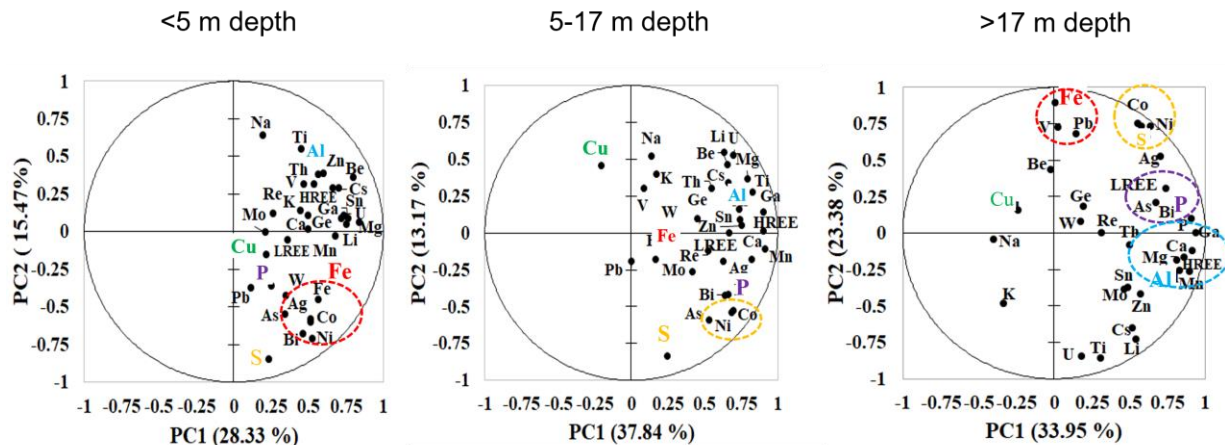
574 *4.4 Multivariate Statistical Analyses*

575
576 Total digestion geochemistry results were studied using the PCA statistical method to define
577 elemental and mineralogical associations as well as to observe possible weathering/ alteration/
578 depositional processes. Samples were split into the HREE groups (< 5 m, 5-17 m and > 17 m
579 depth) and submitted to the PCA study. This differentiation was made based on the trends of

580 the concentrations obtained by total digestion of Cu, Co, S and Fe, and changes in the
581 mineralogical composition (pyrite, apatite and calcite) along the depth profile (Fig. S10 and Fig.
582 2). As can be observed in Fig. 5 (left panel) 43.8 % of the variance of the samples from 0 to 5 m
583 depth are explained by two components. The first principal component (PC1) is by far the
584 predominant one and accounts for 28.33 % of the variance, while the second one (PC2) only
585 explains 15.47 %. PC1 is related to silicate minerals, mainly aluminosilicate of Mg, Ca, and Mn,
586 as well as to immobile elements like Cs, Ga, Be, Li, Sn, U, Ag, Co, Ni, Th, Zn and HREE. A
587 Pearson's correlation matrix (Fig. S13) shows that Al exhibits a good positive correlation with
588 HREE (0.651) but no correlation with LREE (0.319). This could be attributed to the adsorption of
589 HREE on micas under the neutral or low pH conditions of the tailings deposit. Yang et al.,
590 (2019) indicate that at low ionic strength the adsorption of REEs on kaolinite and halloysite
591 increases with increasing pH conditions. The second principal component could be attributed to
592 the higher presence of sulfide minerals in these surficial samples, as shown by the high loading
593 shown for S (-0.844). The cluster of elements close to Fe (i.e., As, Bi, Co and Ni) are clearly
594 related to the presence of pyrite in the deposit of tailings. A Pearson's correlation matrix shows
595 that S exhibits a strong positive correlation with Ni (0.709) and Co (0.817) and a good positive
596 correlation (> 0.5) with Fe, Ag, As, and Bi, supporting the previous interpretation (Fig. S13).
597 Pyrite is the most abundant sulfide in the deposit of tailings, a mineral that can incorporate these
598 elements within its crystal structure (Abraitis et al., 2004; Moncur et al., 2005; Paktunc et al.,
599 2006; Deditius et al., 2011).

600

601



602

603 Fig. 5. Principal Component Analyses for tailings samples with depths < 5 m, between 5-17 m and > 17
604 m.

605

606 The PCA from samples with depths in the range of 5-17 m are shown in Fig. 5 second panel.
607 As can be observed, 51.01 % of the sample's variance can be represented by two components
608 (PC1 37.84 % and PC2 13.17 %). As in the most surficial samples, the first principal component
609 could be related to Mg, Ca, and Mn aluminum-silicate minerals, as well as associated to
610 immobile elements whereas the second main component could be attributed to the presence of
611 sulfide minerals, mainly pyrite. However, the elements clustering and the interpretations
612 assigned to this second set of samples are not as evident and robust as for the most surficial
613 samples. Finally, the PCA from samples with depths higher than 17 m are shown in Fig. 5 third
614 panel. As can be observed, 57.3 % of variance can be represented by two components (PC1
615 33.95 % and PC2 23.38 %). At least four different element clusters were identified in this PCA.
616 The first principal component is associated with elements that cluster around P and LREE
617 (attributed to the presence of apatite in the tailings) and a cluster that associates with Al (Ca,
618 Mg, Mn and HREE) that likely reflect the silicate minerals in the tailings. A Pearson's correlation
619 matrix shows that P exhibits a strong positive correlation with HREE (0.800) and LREE (0.697)

620 (Fig. S13). However, HREE also show a very good Pearson's correlation with Al (0.834).
621 Therefore, in the deeper samples the HREE and LREE could be incorporated into the crystalline
622 structure of apatite as well as in other silicate minerals. The association of elements in the
623 second principal component determines a cluster around S (Ni and Co), most probably
624 reflecting the presence of pyrite. A Pearson's correlation matrix (Fig. S13) shows that S exhibits
625 a strong positive correlation with Co (0.921), Ni (0.919) as well as with Ag (0.824). A second
626 cluster around Fe is also evident for PC2, with V and Pb showing high positive loadings.
627 According to a Pearson's correlation matrix, Fe exhibits a strong positive correlation with V
628 (0.740) and a good correlation with Pb (0.507). This cluster can be attributed to the presence of
629 magnetite and hematite within the tailings deposit. The positive correlation with Pb and V could
630 be attributed to the adsorption of these metals on Fe (III) oxides (Liang et al., 2017).

631
632 Based on the main elements grouped by the PCA, one-way ANOVA and Welch test analyses
633 were carried out in order to determine significant differences among the element concentrations
634 in the three studied depth ranges (i.e., < 5 m, 5-17 m and > 17 m; Fig. S11). The results from
635 this study are clearly in good agreement with previous mineralogical and geochemical
636 observations:

- 637
- 638 1) Al (linked to the presence of aluminosilicates) and HREE concentrations are significantly
639 lower in the deeper levels (> 17 m);
 - 640 2) P concentrations are significantly lower in surficial samples (< 5 m depth), however, the
641 lowest concentrations of LREE correspond to the samples located between 5-17 m depth;
 - 642 3) S and Cu concentrations are significantly higher in the deeper levels (> 17 m depth) in
643 accordance with the higher relative abundance of pyrite in deeper samples;

644 4) Fe (and V) concentrations do not show any significant difference among the three depth
645 groups, as it would be expected from the homogeneous distribution of the major oxides
646 (magnetite and hematite, Fig. 2) in the deposit of tailings.

647 One way ANOVA and Welch test analysis were also used to determine statistically significant
648 differences among the horizontal elemental concentrations. To this end, the samples were
649 grouped according to their presence in the eastern, central, and western sectors of the deposit
650 of tailings (Fig. S12). The results can be summarized as follows: 1) Al, REEs, P and Ca
651 concentrations are significantly higher in samples located in the western sector; 2) Fe, Cu, and
652 Pb concentrations are significantly higher in samples located in the eastern sector. The
653 concentrations of the second group of elements increase in the opposite direction than the first
654 group. These trends are in accordance with the mineralogical characterization, and they could
655 be attributed to a gravitational segregation process during deposition and accumulation. As it
656 occurs in most mine tailings deposits (Pan et al., 2014) minerals with higher specific gravities
657 (like sulfides and Fe-oxides) tend to settle down in the area closest to the dam wall (where the
658 tailings discharge is produced) whereas minerals with lower specific gravities (like
659 aluminosilicates) typically reach the most distal areas of the deposit.

660

661 *4.5 Sequential extraction*

662

663 A mineral sequential extraction procedure (SEP) was implemented as an additional tool
664 linking the observed mineralogy and bulk chemistry and to generate a better understanding of
665 elemental mobility within the tailings profile. The methodology of seven steps developed by Dold
666 and Fontboté (2001) was used. This procedure was optimized for IOCG-type tailings deposits
667 located in the vicinity of the study area and with similar mineralogical composition to El Buitre
668 tailings deposit.

669 The SEP results obtained from three selected boreholes (i.e., B-04, B-13, and AD-02) were
670 averaged to have a representative value for each depth (Table 3). Taking into consideration the
671 mineralogical and geochemical information previously presented, as well as the multivariate
672 statistical analysis, the following elements were selected to study the selective dissolution of the
673 most relevant minerals in the tailings deposit: 1) Al and Mg, as indicators of primary silicate
674 minerals; 2) Ca, as indicator of carbonates and gypsum; 3) S, as indicator of sulfide and sulfate
675 minerals, 4) Cu, as indicators of chalcopyrite and 5) Fe, as indicator of Fe (III) oxides, Fe (III)
676 oxyhydroxide and sulfides. Other elements were also studied to characterize the mobility of
677 potential pollutants (i.e., As, Mn, Cr and Ni) and critical raw elements (i.e., V and Co) within the
678 tailings deposit.

679
680 The aluminosilicates (marked by the combined presence of Al and Mg) are mostly dissolved
681 in the very last step of the SEP (i.e., Step 7: residual fraction), and they did not show any
682 discernible trend along the depth profile. Gypsum distribution, marked by Ca and S release
683 during the first step of the SEP (Ca/S molar = 1 in Step 1), did not show significant differences
684 through the depth profile (Table 3). In addition, the saturation index calculated using PHREEQC
685 corroborates the precipitation of gypsum by saturation, product of pyrite and calcite dissolution
686 in oxidizing conditions (Table S7). Calcite dissolution can be traced by Ca release, mostly
687 limited to the second step of the SEP (exchangeable fraction and carbonates). As shown in Fig.
688 6, calcite concentration progressively increases with depth. Mn is also predominantly released
689 during Step 2, following a similar distribution as the one described for Ca. In addition, a Pearson
690 'correlation (Fig. S13) shows a strong correlation between Mn and Ca for samples <5 m (0.803),
691 in the range from 5-17 m (0.909) and >17 m (0.915). Regarding the sulfide minerals, it can be
692 observed how sulfur is predominantly released in step 5, and to a lesser extent in step 6, as
693 opposed to what could be expected from the optical mineralogical study that presented
694 secondary Cu sulfide as a minor sulfide and pyrite as major sulfide phase. Dold and Fontboté

695 (2001) already observed this type of inconsistencies in other mine tailings and they attributed it
696 to the presence of unstable pyrite minerals from the oxidation zone that may partially dissolved
697 in the fifth step of the SEP. Notwithstanding, almost no S reached Step 7, so sulfide selective
698 dissolution was achieved between Steps 5 and 6. Combining Steps 5 and 6, it can be observed
699 how sulfide concentrations slightly increases with depth (Table 3), in accordance with previous
700 mineralogical observations (Fig. 2). Cu concentration (adding all the steps) decreases from the
701 surface to the center of the tailings deposit to finally increase again in the deepest levels, in
702 accordance with previous observations (Fig. 4). Finally, Fe (III) oxyhydroxides and Fe (III)
703 oxides should be selectively dissolved in the third and fourth step of the SEP. However, most of
704 the Fe and its associated elements (V and Cr) were predominantly released (Table 3 and S6) in
705 the last step of the SEP (residual fraction). This discrepancy could be explained because the
706 solubility of Fe oxides is highly dependent on crystallinity. Highly crystalline Fe (III) oxides
707 remain practically unchanged (and their Fe is not released) until the last step of the SEP
708 (Caraballo et al., 2018).

709
710 According to the Chilean Norm 46/2002 for the emission of liquid waste into groundwater in
711 highly vulnerable environments, the concentrations of As and Ni were found in low
712 concentrations (Ministerio Secretaría General de la Presidencia de Chile, 2002). In addition,
713 these elements remain mostly immobile until the last steps of the SEP (Table S6), when pyrite is
714 dissolved. On the other hand, Mn is more mobile, and it is significantly released in step 2, during
715 the neutralization reactions induced by calcite dissolution (Table S6). Finally, V and Cr show
716 very limited mobility (Table S6) that is coupled to the dissolution of the Fe oxides (i.e., hematite
717 and magnetite).

718

719 **5. Conceptual model**

720 In the deposit of tailings it is possible to distinguish geochemical and mineralogical variations
721 with increasing depth and in the area distal to the wall dam or the waste discharge zone. The
722 presence of multiple discharge points and the geometry of the deposit, favor the stratification of
723 the tailings due to coarse particle segregation and settling near the spigotting points.
724 Additionally, tailings stratification may be associated with mineralogical changes, and process
725 parameters such as variation grinding size during the production period of the deposit of tailings
726 (Mulenshi et al., 2019, 2021; Pan et al., 2014). Therefore, if certain minerals occur in a particular
727 particle size, they may segregate and accumulate in specific zones of the deposit (Lottermoser,
728 2010). This trend was evidenced in section 4.4, where Al, HREE, P, Ca, Fe, Cu and Pb
729 concentrations presented lateral stratifications. Elements such as Fe, Cu, and Pb could have
730 been associated with coarse particles and tended to settle near discharge points. While Al,
731 HREE, P and Ca could have been associated with finer and lighter particle sizes and were
732 settling further away.

733
734 On the other hand, in order to delimit the possible alteration fronts along the depth profile of
735 the deposit of tailings, the distribution patterns of pyrite, chalcopyrite and calcite were analyzed,
736 as they represent the most reactive minerals. In addition, we analyzed the Co (representative of
737 the trends of Cu, S and Fe) and LREE (representative of the trends of REEs) concentrations
738 obtained by total digestion, the geostatistical distribution model, and the results of the different
739 sequential extraction steps. An oxidation zone with an extension between 3-5 m in depth was
740 determined, characterized by a lesser relative abundance of pyrite, chalcopyrite, and calcite with
741 respect to the deeper levels. In this layer, the sulfide minerals present low degrees of alteration
742 and the absence of Fe (III) oxyhydroxides rims or coatings. Below the oxidation zone, the
743 sulfides remain unchanged, with constant relative abundance percentages, up to the depth

744 range between 17-18 m. This layer was defined as the primary zone. Subsequently, towards
745 the deeper layers (> 17 m) a significant increase in pyrite and S concentrations occur, which are
746 reflected in the one-way ANOVA and Welch test analysis performed in section 4.4, and in steps
747 5 and 6 of the SEP. In addition, elements associated with pyrite such as Co, Cu and Fe, and
748 elements associated with apatite such as P, are slightly enriched at the bottom of the deposit of
749 tailings, mainly in the depth range 30 – 40 m (Fig. 4 and Fig. S6). This last layer was present in
750 the deepest sectors, specifically in the central and eastern sectors of the deposit of tailings
751 (Fig.7). It was considered as a primary zone enriched in pyrite and apatite. The highest
752 proportion of these minerals in depth was attributed to the combination of three factors: 1) the
753 tailings dam geometry and the gravitational deposition of heavier mineral particles in the tailings
754 slurry that promote the relative enrichment of pyrite and apatite in the deep zone; 2) limited
755 oxygen diffusion at the bottom of the deposit of tailings. Therefore, the biotic oxidation reactions
756 are dominant, and they could be restricted by the low availability of Fe (III) in solution, due to
757 neutral pH conditions and low water flow, preserving pyrite; 3) there is little information about
758 the mine that gave rise to the mine wastes, so it is possible that the older tailings represent a
759 more pyrite-enriched portion of the ore deposit respect more recent materials, or could even
760 correspond to materials from a different mine nearby the flotation plant.

761

762 According to the conceptual model proposed in Fig. 7, it is observed that the concentrations
763 of the LREE (present in greater proportion than the HREE) represented in the section of the
764 central-eastern area of the deposit of tailings, do not present enrichment in the areas deeper of
765 the deposit. This trend corresponds to the observations made in the geostatistical model and it
766 is probably associated with the double association of REEs with aluminosilicates and apatite
767 minerals.

768

769 Regarding pH conditions within the tailings deposit, paste pH values remain relatively
770 constant (around 7.9) along the depth profile which indicates a relative homogeneous
771 distribution of calcite. The neutral pH values, low water content of the mine residue and high
772 evaporation in arid climates diminish the mobility of the elements liberated from the oxidation
773 zone and limit the oxidation reactions to fine-grained horizons (relatively poor in sulfide
774 minerals) due to their higher water retention capacity (Dold and Fontboté, 2002). In addition, the
775 low-sulfides characteristic of tailings from IOCG reservoirs together with the restriction of the
776 mobility of their oxidation products, explains the absence of secondary efflorescent salts on the
777 surface and the development of a secondary mineral enrichment zone. The above indicates that
778 the deposit of tailings is environmentally safe, due to the restricted mobility of potential polluting
779 elements.

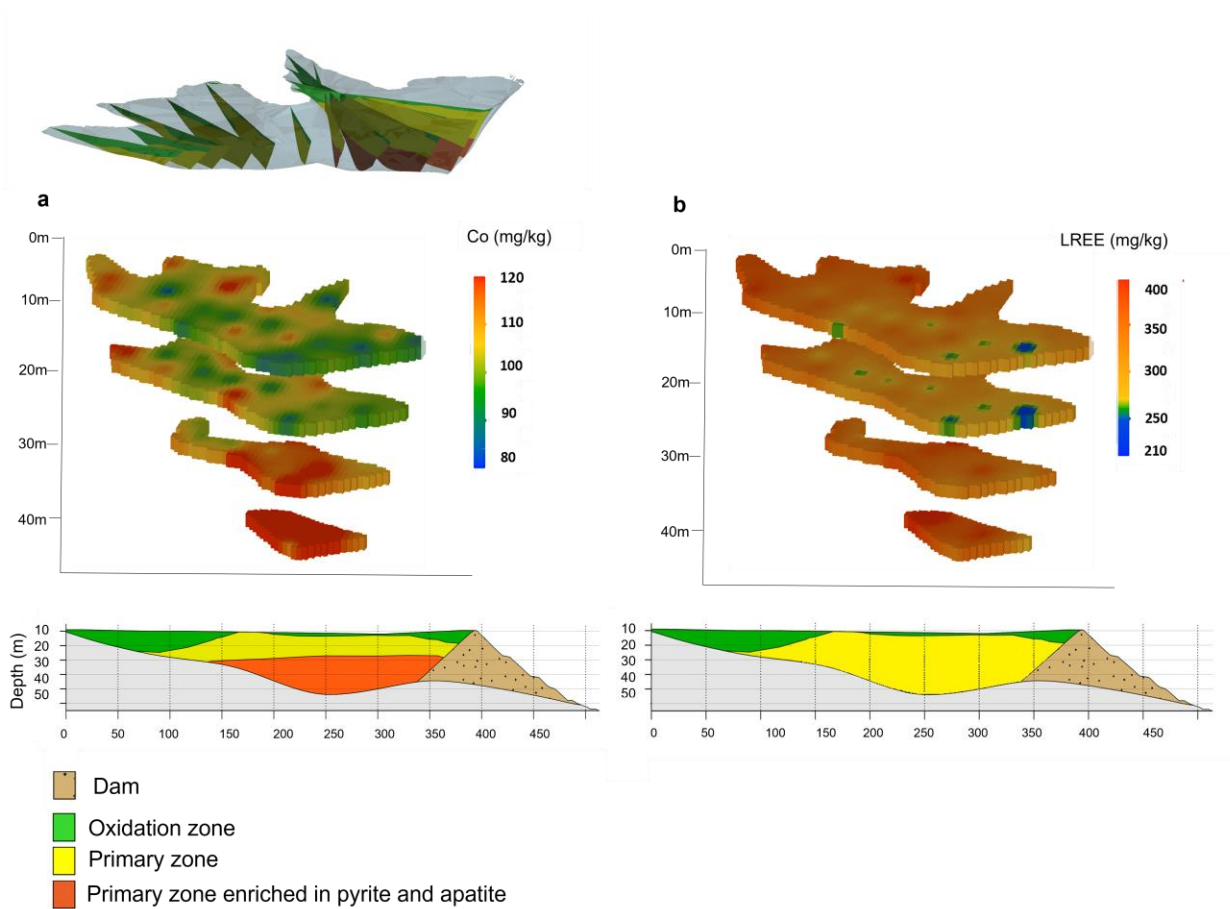
780

781 On the other hand, some important aspects to highlight for future revalorization studies are: 1.
782 The economic recovery of the REEs from the reprocessing of the tailings is improbable,
783 because they are associated with both aluminosilicates and apatite minerals. Which makes it
784 difficult to concentrate these elements as by-products. 2. The high content of magnetite could
785 allow economic recovery of Fe by magnetic separation. This method is currently being applied
786 profitably by CAP's subsidiary Compañía Minera del Pacífico (CMP), located near of El Buitre
787 tailings deposit. CMP produces pellet feed with tailings from the copper concentrator plant of
788 Compañía Contractual Minera Candelaria, with average Fe grades of 11.4 % (Compañía Minera
789 del Pacífico, 2020). 3. The elimination of impurities from the Fe concentrate could favor the
790 concentration of Cu and its possible economic recovery. In addition, it would allow to
791 concentrate pyrite and recover the Co associated with its crystalline structure.

792

793 Taking into consideration all the above mentioned characteristics, a 3D conceptual model
794 was drawn using Leapfrog. To facilitate the model visualization, it was decided to highlight

795 several representative 2D sections (Fig. 7a) of the Co concentrations along the deposit of
796 tailings. Additionally, two detailed representative section of the Co and LREE concentrations of
797 the central-eastern area of the deposit of tailings are shown. These sections were contrasted
798 with the geostatistical modeling.



799

800 Fig. 7. a. Modeled Co concentration and cross-section of the distribution of the alteration fronts in the
801 central-eastern area of the deposit of tailings. Additionally, the upper left figure shows different sections
802 along the geometry of the deposit of tailings, indicating the absence of the layer enriched in sulfides and
803 apatite, in the western sector of the deposit. b. Modeled LREE concentration and cross-section of the
804 distribution of the alteration fronts in the central-eastern area of the deposit of tailings. This trend is
805 representative of the distribution of REEs in the deposit of tailings.

806

807 **6. Conclusions, implications, and future challenges**

808 The following conclusions and implications can be extracted from the present study:

809

810 The low water flow in arid climates and the mobility restrictions of metals under neutral pore
811 water pHs, inhibit the development of extensive oxidation fronts and mineral enrichment zones
812 and prevent the appearance of efflorescent salts precipitated on the tailings surface. Therefore,
813 the distribution of chemical elements is mainly controlled by the source deposit mineral
814 composition, the mineral processing technology, gravitational deposition processes and the
815 geometry of the deposit of tailings.

816

817 Due to the low mobility of potential polluting elements, the tailings dam presents a low risk of
818 groundwater contamination, if the current environmental conditions remain.

819

820 The depositional history of this tailings deposit during its operation has shown the generation
821 of layers with different geochemical and mineralogical characteristics (found at the bottom of
822 this tailings deposit), that are unaffected by the limited supergenic weathering processes. These
823 differences are most probably due to a different origin of the ore material processed in the
824 mineral processing plant, whether it comes from a different section of the ore deposit with
825 slightly different mineralogical characteristics (hydrothermal alteration zones, veins, etc.) or from
826 a different mine located in the vicinity of the deposit of tailings.

827

828 Despite the presence of slight stratifications within the tailings deposit, from a reprocessing
829 perspective, the geochemistry and mineralogy of the tailings is quite homogeneous, and it
830 exhibits similar grades for potentially valuable elements (i.e., Fe, Cu, and Co) along the depth
831 and horizontal profiles. Simplifying its exploration and exploitation for future revalorization

832 studies that could focus on the recovery of Fe by magnetic separation. This process could favor
833 the concentration of Cu and Co in the purification stages of the magnetic concentrate. In
834 contrast, the recovery of REEs as a by-product is improbable due to their double association,
835 with apatite and aluminosilicate minerals.

836

837 The following future challenges are envisioned as a result of this study:

838 The relative homogeneity of the grades within the tailings deposit opens the possibility of
839 restructuring the initial sampling mesh. Geostatistical studies are needed to evaluate the
840 possibility of decreasing the number and density of boreholes while maintaining a robust and
841 representative characterization of the deposit. This could drastically reduce the exploration cost
842 of this marginally profitable secondary deposit.

843

844 Also, the present study has shown how limited the weathering and enrichment/depletion
845 processes are in this type of neutral paste-pH tailings deposits in an arid climate. However, this
846 slow mineral weathering cannot be neglected and robust hydrogeochemical models are needed
847 to better understand the formation and evolution of these deposit of tailings stored (or to be
848 stored) for decades. This information is essential to anticipate their long-term chemical stability
849 and/or their possible generation of bigger and deeper oxidation zones within older deposits. To
850 this end, another future challenge is to decipher the role played by microbial communities in the
851 biogeochemical processes involved in the deposit of tailings weathering and evolution in time.

852

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856

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858 All authors read and approved the final manuscript.

859

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869 **10. References**

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	Min.	Max.	Mean	Median	SD	Percentiles			
						25.0	50.0	75.0	95.0
Major									
Fe	9	26	16	15	2	14	15	17	20
Al	1	10	5	5	2	4	5	6	8
Ca	2	6	3	3	1	2	3	3	4
Mg	1	5	2	2	1	2	2	3	3
Si* (XRF)	0.2	3	2	2	0.4	1	2	2	2
K	0.7	5	2	2	0.4	2	2	3	3
S	0.2	3	2	2	0.4	1	2	2	2
Na	0.3	2	0.9	0.9	0.3	0.7	0.9	1	2
Ti	0.1	1	0.3	0.3	0.1	0.2	0.3	0.3	0.3
P	0.1	0.2	0.1	0.1	0.0	0.1	0.1	0.1	0.2
Minor									
Cu	227	4290	795	647	514	433	647	990	1790
LREE	67	800	315	306	110	234	306	373	520
V	62	579	108	105	29	91	106	121	150
Co	25	303	103	100	32	83	100	120	158
Trace									
Ni	28	178	79	76	22	63	76	93	121
Zn	35	196	63	60	18	50	60	71	93
As	1	107	38	37	12	30	37	43	59
Li	14	79	30	26	10	22	26	33	52
Ga	14	41	24	25	5	19	25	28	32
HREE	10	44	22	22	5	18	22	25	30
Pb	2	38	12	11	6	8	11	16	24
W	0	101	7	6	6	4	6	8	12
Th	1	9	3	3	1	3	3	4	5
Sn	1	18	3	2	2	2	2	3	5
Cs	1	7	2	2	1	1	2	2	3
U	1	3	2	2	0.3	2	2	2	2
Ag	0.1	4	1	1	1	1	1	1	2
Mo	0.1	4	1	1	1	1	1	1	2
Be	1	3	1	1	0.3	1	1	1	2
Ge	0.1	2	1	1	0.4	0.3	1	1	1
Bi	0.1	4	1	1	0.3	0.4	1	1	1

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1150 Table 1. Univariate statistics for selected elements of the tailings with Min., Max., Mean, Median,
1151 standard deviation (SD) and percentiles (25, 50, 75 and 95). Major elements in % wt and minor
1152 and trace elements in mg/kg. The concentrations correspond to the total digestion analysis (N=
1153 755 samples). The concentration of Cd, Ba, Cr, Nb, Sb and Se were below the detection limit.

Elements	Cut-off	Mean grade	Percentage (%)	Reserves (Mt)	Metal prices ^a (US\$/ton)	Economic potential (MUS\$)
REEs	200 mg/kg	337 mg/kg	100	0.0022	4651315	10232.9
Co	190 mg/kg	103 mg/kg	0	0	72780	-
Cu	0.10 wt %	0.07 wt %	2.1	0.00009	9929	0.9
Fe	11.4 wt %	15.50 wt %	100	1.023	141.1	144345.3

1154 ^a Sources: (LME, 2022), Daily metal prices (2022) and Mineralprices.com (2022)

1155 Table 2. Rough estimation of potential resources of El Buitre tailings deposit

Steps	1	2	3	4	5	6	7
Aluminum	Mean \pm SD						
Depth (m)							Silicates
1	<DL	18 \pm 5	54 \pm 5	43 \pm 2	12 \pm 1	246 \pm 12	1633 \pm 10 a
3	<DL	26 \pm 5	64 \pm 3	44 \pm 3	14 \pm 0	233 \pm 10	1620 \pm 3 a
9	<DL	18 \pm 5	55 \pm 11	45 \pm 10	13 \pm 3	259 \pm 82	1415 \pm 162 a
15	<DL	17 \pm 1	68 \pm 2	52 \pm 1	15 \pm 0	315 \pm 62	1612 \pm 37 a
21	<DL	18 \pm 4	68 \pm 5	45 \pm 7	15 \pm 1	310 \pm 38	1456 \pm 38 a
35	<DL	19 \pm 0	82 \pm 11	57 \pm 2	20 \pm 5	408 \pm 48	1425 \pm 100 a
Calcium	Gypsum	Carbonates					
1	137 \pm 31 a	297 \pm 84 a	<DL	<DL	52 \pm 5	21 \pm 0	53 \pm 44
3	108 \pm 32 a	411 \pm 114 ab	<DL	<DL	69 \pm 11	23 \pm 5	22 \pm 4
9	107 \pm 16 a	564 \pm 26 b	<DL	<DL	91 \pm 15	34 \pm 5	<DL
15	99 \pm 1 a	616 \pm 43 b	<DL	<DL	109 \pm 6	31 \pm 1	<DL
21	138 \pm 69 a	625 \pm 61 b	<DL	<DL	105 \pm 3	33 \pm 1	19 \pm 6
35	128 \pm 25 a	806 \pm 83 c	<DL	<DL	140 \pm 24	38 \pm 1	<DL
Sulphur	Gypsum				Secondary sulfide	Primary sulfide	
1	134 \pm 33 a	<DL	<DL	<DL	274 \pm 7 a	68 \pm 12 a	<DL
3	120 \pm 7 a	<DL	<DL	<DL	324 \pm 33 ab	146 \pm 5 b	40 \pm 6
9	83 \pm 5 a	<DL	<DL	<DL	404 \pm 27 ab	137 \pm 69 b	33 \pm 1
15	96 \pm 1 a	<DL	<DL	<DL	418 \pm 14 ab	78 \pm 7 a	<DL
21	140 \pm 76 a	<DL	<DL	<DL	432 \pm 16 ab	159 \pm 18 b	16 \pm 9
35	128 \pm 27 a	<DL	<DL	<DL	497 \pm 13 b	164 \pm 62 b	12 \pm 1
Copper					Secondary sulfide	Primary sulfide	
1	<DL	6 \pm 1	5 \pm 1	0.6 \pm 0	5 \pm 0.3 a	1 \pm 0.1 a	0.3 \pm 0.1
3	<DL	4 \pm 0	2 \pm 0	0.3 \pm 0	10 \pm 2 b	2 \pm 1 a	0.3 \pm 0
9	<DL	2 \pm 0	2 \pm 0	0.3 \pm 0	4 \pm 0.1 a	1 \pm 0.1 a	0.3 \pm 0.1
15	<DL	2 \pm 0	1 \pm 0	0.3 \pm 0	3 \pm 0.1 a	0.7 \pm 0.1 a	0.2 \pm 0
21	<DL	2 \pm 0	2 \pm 1	0.3 \pm 0	5 \pm 1 a	1 \pm 0.1 a	0.1 \pm 0
35	<DL	5 \pm 1	3 \pm 1	0.5 \pm 0	13 \pm 0.1 b	2 \pm 0 b	0.3 \pm 0.1
Iron			Fe (III)	Fe (III) Oxides		Silicates	
			Oxyhydroxides				
1	<DL	31 \pm 9	289 \pm 24 a	178 \pm 24 a	74 \pm 1	538 \pm 10	1637 \pm 3
3	<DL	44 \pm 6	258 \pm 2 a	192 \pm 9 a	85 \pm 3	583 \pm 13	1716 \pm 188
9	<DL	38 \pm 8	291 \pm 10 a	195 \pm 19 a	116 \pm 17	626 \pm 60	1902 \pm 139
15	<DL	43 \pm 1	270 \pm 18 a	193 \pm 18 a	136 \pm 14	664 \pm 84	1492 \pm 216
21	<DL	38 \pm 9	320 \pm 79 a	201 \pm 11 a	127 \pm 15	706 \pm 103	1570 \pm 234
35	<DL	41 \pm 1	302 \pm 17 a	206 \pm 18 a	151 \pm 18	825 \pm 38	1155 \pm 0

Table 3. Results of the sequential extraction corresponding to the average of boreholes Ad2, B-04 and B-13. The values are expressed in mmol/kg. Elements below the detection limit are indicated as <DL. Step 1: Water soluble fraction. Step 2: Exchangeable fraction, Step 3: Fe (III) oxyhydroxides, Step 4: Fe (III) oxides, Step 5: Organics and secondary Cu-sulfides, Step 6: Primary sulfides and Step 7: Residual fraction. Different letters indicate that there is a significant difference between the concentration of the elements at a certain depth. On the contrary, the same letters indicate that there are no significant differences.