

1           EFFECT OF BLEND MIXING AND FORMULATION ON  
2           THERMOPHYSICAL PROPERTIES OF GLUTEN-BASED  
3           PLASTICS

4 ESTEFANÍA ÁLVAREZ-CASTILLO <sup>1</sup> ✉, MARÍA RAMOS <sup>1</sup>, CARLOS BENGOCHEA <sup>1</sup>, INMACULADA  
5 MARTÍNEZ <sup>2</sup>, ALBERTO ROMERO <sup>3</sup>

6 *<sup>1</sup> Departamento de Ingeniería Química, Universidad de Sevilla, Escuela Politécnica Superior,  
7 Calle Virgen de África 7, 41011, Sevilla, Spain*

8 *<sup>2</sup> Departamento de Ingeniería Química, Centro de Investigación en Tecnología de Productos y  
9 Procesos Químicos (Pro2TecS), Campus de Excelencia Internacional Agroalimentario, ceiA3,  
10 Universidad de Huelva, Campus El Carmen, 21071 Huelva, Spain*

11 *<sup>3</sup> Departamento de Ingeniería Química, Facultad de Química, Universidad de Sevilla, 41012,  
12 Sevilla, Spain*

13  
14 ✉ MALVAREZ43@US.ES

15 Departamento de Ingeniería Química, Universidad de Sevilla

16

17

18

19

20

21

## 22 **Abstract**

23       Gluten-based plastics represent an interesting biodegradable alternative to conventional  
24 plastics, due to their greater mechanical properties when compared to other protein-based  
25 materials. In this manuscript, the effect that both blend mixing and the presence of sugars exerts  
26 on the gluten-based material properties was evaluated through rheological assays, water immersion  
27 tests, and microscopy. Thus, two different mixing procedures were performed (extrusion and  
28 internal mixing) for gluten samples plasticized by water and glycerol that were eventually injection  
29 moulded. The effect of trehalose and sucrose was studied including a 20 wt.% content in the  
30 material formulation. It was observed that, even if no great differences were noticed when using  
31 different mixing processes, extruded blends, as well as the resulting bioplastics, displayed higher  
32 viscoelastic moduli than those mixed using an internal mixer. For both mixing procedures,  
33 bioplastics displayed a much more pronounced thermoplastic behaviour when compared to blends.  
34 The plasticising effect of the sugar in the gluten-based plastics was apparent, denoted by a decrease  
35 in the viscoelastic properties, together with an increase in the water uptake capacity and their  
36 porosity. Interestingly, a superabsorbent material could be obtained when including trehalose in  
37 extruded samples, showing the feasibility of using gluten plastics in this field.

38 **Keywords:** *gluten; sucrose; trehalose; plastic*

39

## 40 **1. Introduction**

41 There is an increasing interest in replacing common synthetic polymers with more  
42 environmentally friendly materials, which are commonly composed by organic materials, such as  
43 by-products of the food industry (Álvarez-Castillo et al., 2020a). In this sense, polysaccharides  
44 (e.g., starch and cellulose derivatives) and proteins (e.g., soy, pea, plasma, or wheat gluten) have  
45 been widely used in the manufacture of green plastics (Álvarez-Castillo et al., 2020a, 2018; Jerez  
46 et al., 2005).

47 Gluten is a protein that may be obtained from wheat, a cereal that started to be grown about  
48 10,000 years ago in Southeast Asia. Wheat is one of the most important cereals along with corn  
49 and rice, with an annual production of 750.4 Mt in 2016-2017, occupying some 221.6 million  
50 hectares worldwide, according to the Foreign Agricultural Service (USDA, 2018). This production  
51 decreased 2.7% in 2018, when around 736.1 Mt were produced, being Asia the largest producer  
52 (~ 44 %), followed by European Union (~ 20 %) and America (~ 14 %) (Food and Agricultural  
53 Organization of the United Nations, 2019) . Gluten is obtained as a by-product of the starch  
54 industry, the production of which seems to be increasing due to its allergenicity. It contains at least  
55 75% protein by dry weight, being the rest mostly starch and lipids.

56 Gluten is the prolamin fraction of wheat, mainly composed of gliadins and glutenins  
57 (Sothornvit and Krochta, 2005). Gliadins are monomeric proteins, soluble in aqueous alcohols,  
58 with a molecular weight ranging from 30 to 100 kDa, and responsible for the cohesiveness of  
59 gluten. Regarding their structure, four types are distinguished, such as:  $\alpha$ ,  $\beta$ ,  $\gamma$  and  $\omega$ -gliadins.  
60 Instead, glutenins are partially insoluble in most solvents due to their large size, with a molecular  
61 weight around 3000 kDa. This protein fraction seems to be responsible of the elastic properties of

62 gluten (Hoseney and Rogers, 1990), that are explained by the formation non-covalent interactions,  
63 like intra and inter hydrogen bonds of the individual glutenin chains. On the other hand, gliadins  
64 would hinder interactions between glutenin chains, acting as plasticizers that decrease the elasticity  
65 of the dough. Therefore, the ratio of monomeric gliadin to polymeric glutenin would determine  
66 the rheological behaviour of the gluten dough and therefore affects the performance of gluten  
67 protein (Hernandez-Izquierdo and Krochta, 2008).

68 Gluten is widely applied in the food industry, but it should also be highlighted its applications  
69 in the field of plastic materials (Gómez-Martínez et al., 2011; Jerez et al., 2005; Jiménez-Rosado  
70 et al., 2019; Ullsten et al., 2009; Zárate-Ramírez et al., 2011). For the latter purpose, the use of a  
71 plasticizer, such as glycerol or water, is frequently needed to obtain a cohesive and homogeneous  
72 blend that would give rise to plastic materials through common processing techniques.  
73 Additionally, some studies have reported the use of monosaccharides, disaccharides or  
74 oligosaccharides, polyols or even lipids as plasticizers for the manufacture of protein-based  
75 plastics. Sugars have also been reported to generally protect proteins from denaturation, being  
76 trehalose exceptionally superior to most of them (Kaushik and Bhat, 2003). Sucrose and trehalose  
77 are non-reducing disaccharides formed by the union of two monosaccharides by means of an *O*-  
78 glycosidic bond, not possessing any free anomeric carbon. Sucrose is obtained from sugar cane or  
79 sugar beet, while trehalose is mainly obtained from mushrooms, possessing a much lower  
80 sweetening power.

81 Injection moulding is one of the most widely used techniques to obtain protein-based plastic  
82 (Álvarez-Castillo et al., 2018). However, gluten based plastic materials have been mostly  
83 processed by other techniques (e.g. compression, extrusion, casting), not existing many studies  
84 focused on injection-moulded materials (Cho et al., 2011). Considering that a homogeneous blend

85 of the ingredients is a prerequisite for an optimal injection process, blends used to feed the injection  
86 moulding unit were obtained after mixing the raw materials in a prior stage. For this purpose, two  
87 different mixing procedures were compared in the present study: an internal mixing rheometer  
88 system where both temperature and torque can be monitored throughout the mixing stage (Gómez-  
89 Heincke et al., 2016; Zárate-Ramírez et al., 2014a); and a lab-scale twin extruder, which allowed  
90 setting a temperature profile from the hopper to the exit die (Jiménez-Rosado et al., 2019). Both  
91 techniques apply high shear stresses to the blend due to the action of the corresponding dispersing  
92 element (i.e. counter-rotating shafts and endless screw, respectively).

93 As the properties of protein-based plastics depend largely on both the processing conditions  
94 and the formulation (Álvarez-Castillo et al., 2020a, 2019, 2018), the effect of both the mixing  
95 procedure (i.e., internal mixing, extrusion) and the presence of sucrose or trehalose (20 wt. %) was  
96 considered for gluten-based bioplastics. Therefore, the goal of the present study has been to assess  
97 the consequences of different mixing stages, as well as of the presence of sugars, on the rheological  
98 properties, water absorption response and microstructure of gluten-based plastic materials.

## 99 **2. Materials and Methods**

### 100 **2.1 Materials and processing methods**

#### 101 2.1.1. Materials

102 Wheat Gluten (WG), provided by Productos Riba S.A. (Spain), is a protein concentrate  
103 composed of 83 % protein, 3 % lipids, 1 % ash, 8 % humidity, and small amounts of carbohydrates.  
104 Sucrose was supplied by Sigma Aldrich (USA) and trehalose anhydrous by Guangxi Biochemical  
105 Science and Technology (China). Deionized water (W) and glycerol (Gly) supplied by  
106 PANREAC.S.A. (Spain) were used as plasticizers.

107 2.1.2. Processing methods

108 *Blends*

109 In order to obtain homogeneous blends which could be correctly injection moulded, two  
110 different blending strategies were studied.

111 In the first one, a premix was made initially through stirring by hand every fraction (solid  
112 and fluid). Then, solid (gluten, disaccharides) and fluid (glycerol, water) ingredients were  
113 introduced separately in a two-blade counter-rotating batch mixer Haake PolyLab QC  
114 (ThermoHaake, Karlsruhe, Germany) at room temperature (around 25 °C) under adiabatic  
115 conditions. Different rates were tested (50, 100 and 200 rpm) for a certain mixing time (5 and 10  
116 min).

117 In the second method, two feeding units for a twin-screw extruder EuroLab 16  
118 (ThermoScientific, USA) were used: a peristaltic pump for the fluid fraction; and a powdery solid  
119 dispenser, for the addition of wheat gluten. As disaccharides were added in an aqueous solution  
120 onto the extruder, the amount of disaccharide added was limited by the maximum solubility of  
121 trehalose in water at 20 °C (68.9 / 100 g), as the water solubility of sucrose is exceedingly higher  
122 (200 / 100 g). The procedure was as follows: 1) A 60 wt. % disaccharide solution was prepared at  
123 20 °C; 2) then, it was mixed with the proper volume of glycerol; and 3) finally, the plasticizer  
124 fraction was pumped onto the barrel using a peristaltic pump when the extrusion process started.  
125 An adequate selection of the pump rates allowed to achieve the specified formulation (Table 1).  
126 The extruder internal diameter was 19 mm with a length ratio/diameter (L/D) of 40. Temperature  
127 was controlled along the extruder cylinder through ten electrical resistances (50, 55, 50, 50, 50,  
128 50, 50, 50, 50, 60, 60 °C), being the temperature profile defined based on previous work in order  
129 to avoid excessively high temperatures that could promote protein degradation (Fang et al., 2014).

130 The extruder torque per shaft was controlled in order to always remain under the maximum  
 131 allowed value (12 N·m). The chosen outlet head was circular, obtaining extruded polymeric  
 132 filaments that were later cut into pellets suitable for feeding the injection unit.

133 For both procedures, the protein / (water + glycerol) ratio was always kept at 2/1, containing  
 134 the same amount of water and glycerol. This ratio was kept constant when sugar was added. Table  
 135 1 displays the details of each material composition.

136 Table 1. Composition of the systems studied

| System        | Sucrose (%) | Trehalose (%) | Wheat Gluten (%) | Glycerol / Water (%) |
|---------------|-------------|---------------|------------------|----------------------|
| Reference     | -           | -             | 67.00            | 16.50/16.50          |
| 20% sucrose   | 20.00       | -             | 53.30            | 13.35/13.35          |
| 20% trehalose | -           | 20.00         | 53.30            | 13.35/13.35          |

137

138 *Plastic*

139 Blends, obtained in the previous mixing stage either as doughs or extrudates, were further  
 140 processed into rectangular plastic probes (1x10x60 mm) through injection moulding using a  
 141 Minijet II Piston Injection Moulding System (ThermoHaake, Karlsruhe, Germany). Blends were  
 142 first introduced in a cylinder at 80 °C, applying a pressure of 500 bar for 20 s in order to inject the  
 143 samples into a mould at 100 °C, where the samples remained for 180 s under the same pressure.

## 144 **2.2. Characterization methods**

### 145 **2.2.1. Rheological properties**

146 Small amplitude oscillatory measurements were carried out within the linear viscoelastic range in  
 147 compressional and torsional mode to characterize the blends and plastics, respectively. For blends,

148 temperature ramp tests were performed from 25 °C to 140 °C in a RSA3 rheometer (TA  
 149 Instruments, USA), using an 8 mm cylindrical geometry at a heating rate of 5 °C/min and constant  
 150 frequency (1 Hz). For plastics, temperature ramp tests were carried out from -30 °C to 140 °C in a  
 151 DHR-3 rheometer (TA Instruments, USA) employing a rectangular geometry in torsion mode, at  
 152 a heating rate of 5 °C/min and 1 Hz.

### 153 **2.2.2. Water uptake and soluble matter loss**

154 Water uptake capacity (WUC) for gluten-based plastics was determined as in previous studies  
 155 (Álvarez-Castillo et al., 2018): initially, samples were submitted to a dehydrothermal treatment in  
 156 an oven at 50 °C for 24 h ( $w_1$ ); then, they were submitted to an immersion stage in deionised water  
 157 during 24, 48 and 72 h ( $w_2$ ); and, finally, swollen samples were again dried at 50 °C for 24 h ( $w_3$ ).  
 158 WUC and soluble matter loss (SML) values were obtained as indicated in equation 1 and equation  
 159 2:

$$160 \quad WUC (\%) = \frac{w_2 - w_3}{w_3} \cdot 100 \quad (Eq. 1)$$

$$161 \quad SML (\%) = \frac{w_1 - w_3}{w_1} \cdot 100 \quad (Eq. 2)$$

### 162 **2.2.3. Scanning Electron Microscopy**

163 Some swollen samples were transformed onto dried matrices through freeze-drying (-80°C and  
 164 0.01 mbar). Thus, small pieces of them (~ 2.5 mm) were gold coated and examined by Scanning  
 165 Electron Microscopy (SEM), in a ZEISS EVO (USA). The analysis was carried out in the  
 166 microscope with a working distance of 6 mm, a beam current of 11-12 pA, and 10 kV as  
 167 acceleration voltage. The magnification used was 200x. The obtained micrographs were analysed

168 with the public domain software ImageJ, in order to estimate porosity values, considering 10  
169 measurements at least.

### 170 **2.2.3. Statistical analysis**

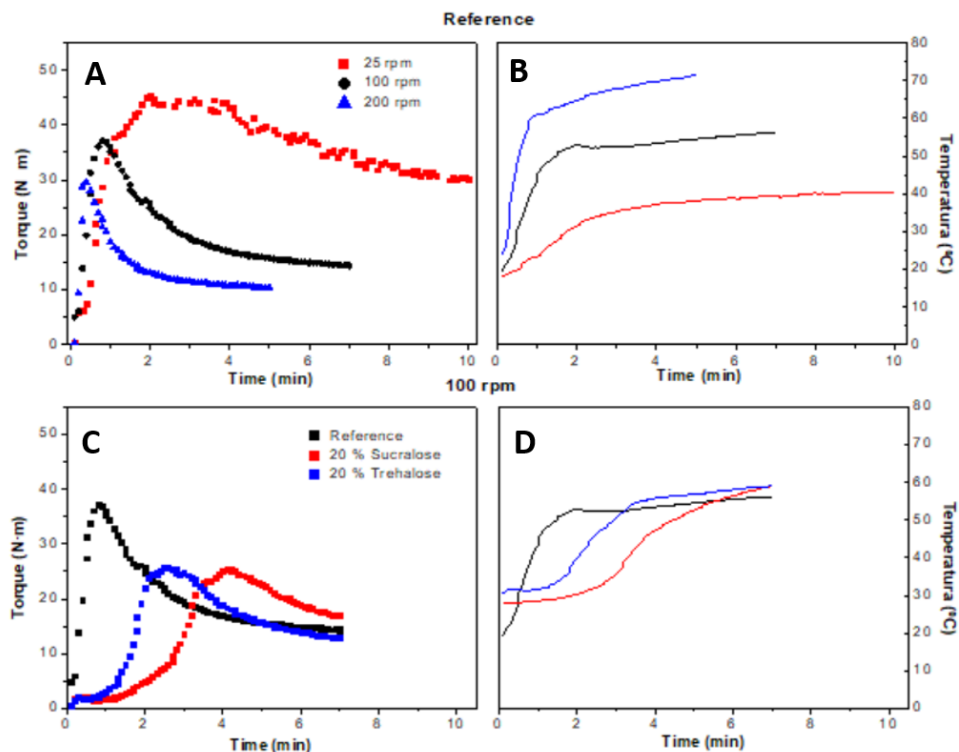
171 Data is presented as mean  $\pm$  standard deviation of three technical repeats using Statsgraphics  
172 software, considering significant probability value of  $p < 0.05$ .

## 173 **3. Results and Discussion**

### 174 **3.1. Effect of the mixing stage**

175 When observing how temperatures evolved along the extruder for the different formulations  
176 studied, it could be observed that temperatures initially matched the set value, but as mixtures went  
177 forward onto the extruder die, a slight deviation towards higher temperatures was detected. This  
178 could be attributed to the dissipation of thermal energy due to the existing friction forces in the  
179 interior of the extruder. Therefore, in the initial section, no deviation was observed, but in the  
180 central section (50 °C as set value), a greater deviation was observed in the reference system (59.8  
181 °C  $\pm$  2.9 °C) than in those containing 20 % of sucrose or trehalose (54.7 °C  $\pm$  1.6 °C, or 56.8 °C  $\pm$   
182 3.6 °C, respectively). Lower relative deviations were found at the exit section (60 °C) either for the  
183 reference or for the samples containing 20 % of sucrose or trehalose (63.5 °C  $\pm$  0.7 °C, 60.5 °C  $\pm$   
184 3.6 °C or 60.7 °C  $\pm$  3.5 °C, respectively). The slightly greater increase of temperature taking place  
185 in the absence of sugar would support the idea of those disaccharides acting as plasticizers, as they  
186 may diminish shear stresses applied to the sample in the extruder due to the relaxation of the  
187 interactions between the polymeric chains exerted by the plasticizers (Ghanbarzadeh et al., 2006).

188        On the other hand, the evolution of torque and temperature with mixing time recorded in the  
189 mixer rheometer for the reference sample systems was illustrated in Figures 1A and 1B for  
190 different mixing rates (25, 100, 200 rpm) and in Figures 1C and 1D for the different formulations  
191 studied (0 % sugar, 20 % sucrose or trehalose) at 100 rpm. When observing the torque evolution,  
192 similar profiles were observed for the reference sample at all mixing rates: at the start of mixing,  
193 the torque suffered a sudden increase up to a maximum value ( $M_{\max}$ ), after which it dropped  
194 continuously until a steady value was achieved. For the reference blend,  $M_{\max}$  values were lower  
195 at higher mixing rates, being located also at shorter times. Thus, when the reference blend was  
196 obtained at 25 rpm,  $M_{\max}$  was achieved after 3 min, being around 1.5 times higher than the  
197 maximum value obtained at 200 rpm after 0.5 min, which could indicate that a more intimate  
198 mixing was previously achieved at higher rates. Mixing conditions have already been reported to  
199 influence largely the properties of blends, finding also similar events when different mixing rates  
200 were used to homogenize WG/Gly blends (Jerez et al., 2005; Redl et al., 1999).



201  
 202 Figure 1. Evolution of torque (A, C) and temperature (B, D) during the mixing process for blends at 25, 100 and 200  
 203 rpm (A, B); and for blends containing 20 % disaccharide (sucrose, trehalose) at 100 rpm (C, D)

204  
 205 Regarding the temperature, an upward evolution was observed during the whole mixing  
 206 process, reaching higher temperatures at the torque peak ( $T_{\text{peak}}$ ) when mixing rates were higher.  
 207 Moreover, this temperature profile also reflected the two regions commented for torque, as  
 208 displayed by the modification of the  $dT/dt$  slope after  $M_{\text{max}}$  was achieved.  $T_{\text{peak}}$  achieved a value  
 209 of 40 °C when the reference was mixed at 25 rpm, while it raised up to 60 °C at 200 rpm. This  
 210 temperature increase may be caused by a higher heat dissipation due to the existence of numerous  
 211 friction forces, which are higher when the mixing rate increased, as have been observed in previous  
 212 studies for WG systems (Jerez et al., 2005; Andrés Redl et al., 1999; Zárate-Ramírez et al., 2011).

213 When sucrose or trehalose was included at 20 wt. % (Figure 1C),  $M_{\text{max}}$  was displaced onto  
 214 longer mixing times, as a lag phase was detected, where no significant increase in the torque

215 occurred. Moreover, the presence of disaccharide resulted in a significant drop of the  $M_{\max}$  values,  
216 which may be attributed to the already commented plasticizing effect of sugars, just like observed  
217 in the temperature profile along the extruder. The torque profile was seemingly independent of the  
218 nature of the disaccharide, as they reached similar values. On the other hand, the temperature  
219 evolution along mixing (Figure 1D) revealed the previously commented correspondence with the  
220 mixing torque, observing, in this case, retardation in the temperature increase when the sugar is  
221 included in the formulation. Anyway, all samples, independently of the formulation, reached  
222 similar  $T_{\max}$  values ( $\sim 60$  °C) at the same mixing rate.

## 223 **3.2. Viscoelastic properties**

### 224 *Blends*

225 The viscoelastic properties of the reference and disaccharide-containing blends obtained either  
226 after extrusion (Figure 2A) or thermoplastic mixing at different rates (Figure 2B, figure 2C and  
227 figure 2D) were assessed carrying out temperature sweep tests in compression mode at a  
228 temperature range from 20 to 120 °C within the LVR. Temperature ramp tests were performed in  
229 order to obtain valuable information that could help to determine the optimal injection conditions  
230 that would facilitate the flow of the blend from the cylinder into the mould. This fact would permit  
231 a complete and homogeneous filling of the mould cavity, which would finally result in a higher  
232 reproducibility in the production of probes (Álvarez-Castillo et al., 2020a; Felix et al., 2015).  
233 Furthermore, from an industrial perspective, a proper study of rheology would imply lower energy  
234 consumption during the injection process. In Figure 2, similar qualitative behaviours were  
235 presented by all samples, independently of the mixing device or mixing rate employed, being the  
236 elastic modulus always higher than the viscous modulus. Thus, a predominantly elastic response  
237 was detected for all blends, typical of solid materials for which both modules show a low

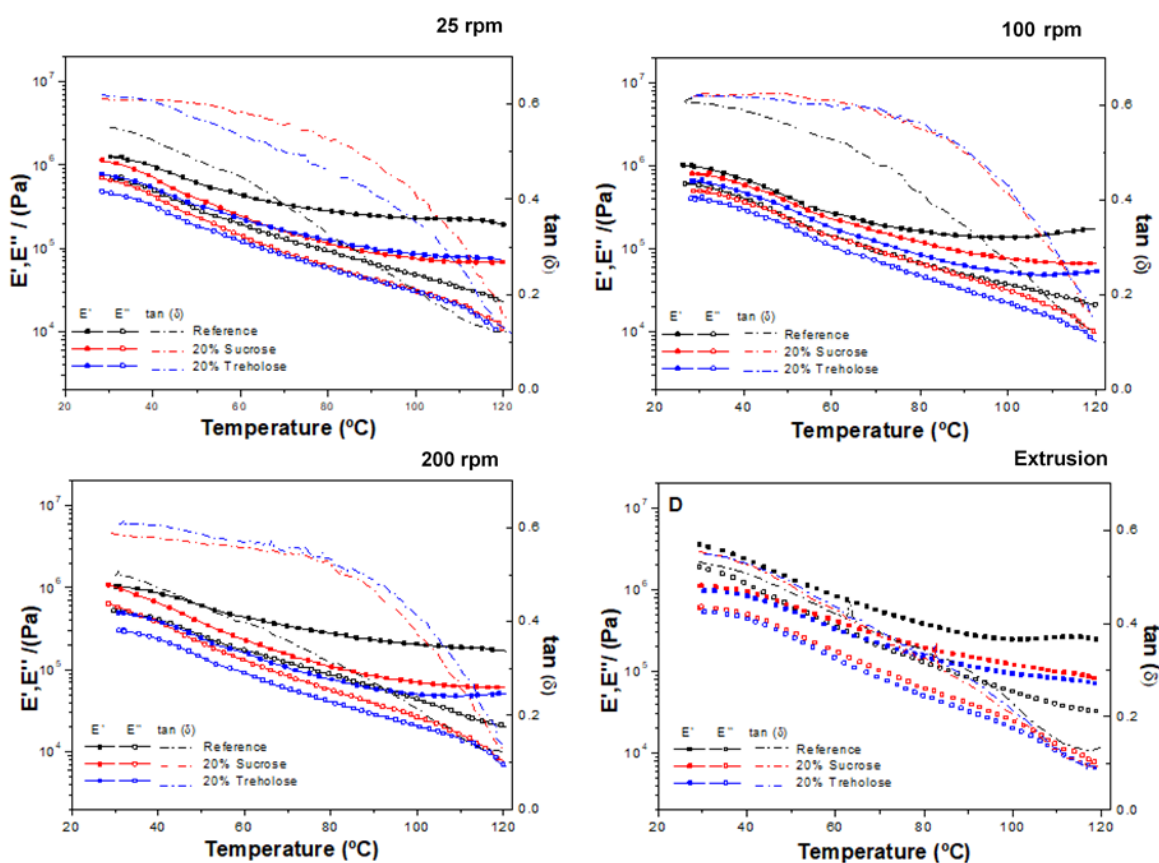
238 dependence with frequency (Álvarez-Castillo et al., 2018; Jerez et al., 2005) (data not shown).  
239 When observing the effect of the mixing device, extruded blends showed slightly higher values  
240 for both elastic and viscous moduli ( $E'$  and  $E''$ , respectively) at room temperature. As identical  
241 raw materials were fed onto both mixing devices, the greater viscoelasticity displayed by extruded  
242 blends could be related to the efficiency of the mixing process. In a continuous twin-screw  
243 extruder, like the one used in the present study, high shear zones are produced at the clearances  
244 either between the screws and the barrel or between the two screws inside the barrel, at which the  
245 flow is ruled by simple shear flow (Emin and Schuchmann, 2013). On the other hand, the  
246 discontinuous mixing device used was a type of internal mixer, consisting of two rotors enclosed  
247 in a mixing chamber, where also high shear forces are generated, dispersing the added ingredients  
248 (Drobny, 2014). In both mixing devices, dispersive and distributive mechanisms must have taken  
249 place. The evaluation of the properties of the blend is helpful to determine the performance of the  
250 mixing process that led to its formation (Kajiwara and Nakayama, 2011). Thus, under the  
251 processing conditions employed, it seems that extrusion resulted in a higher mixing efficiency than  
252 the internal mixer. However, it should be highlighted that the mixing mechanism should not be  
253 inferred from the results obtained, requiring an special assessment of the existing flow patterns  
254 within the mixer, which is a complex issue, beyond the scope of the present manuscript.

255 The addition of a disaccharide, either sucrose or trehalose, in the formulation of these gluten-  
256 based blends prompted a reduction in both viscoelastic moduli during the whole temperature range  
257 studied for both mixing devices. This could be related to the plasticizing effect that the saccharides  
258 seem to possess that relaxes the interactions of the protein chain, allowing greater mobility and  
259 thus lowering the viscoelastic values (Ghanbarzadeh et al., 2006). However, it should be noticed  
260 that the addition of sugar implied a decrease in the gluten, glycerol and water content in the

261 formulation, which may also help to explain the results. However, the gluten/(glycerol+water)  
262 ratio was kept constant in every case (2/1), which may highlight the commented role of plasticizer  
263 exerted by the disaccharide. When observing the effect of temperature, a moderate decrease in  $E'$   
264 occurred initially, generally reaching a rubbery plateau at higher temperatures. On the other hand,  
265  $E''$  displayed a continuous decrease during the whole thermal treatment. Although, as commented,  
266 greater viscoelastic moduli could be found at lower temperatures for the extruded blend, finally,  
267 all blends displayed similar moduli at higher temperatures, independently of the mixing procedure  
268 and speed rate. Those quite alike results when varying rotation speed in the internal mixer was also  
269 observed in similar WG/Gly systems by Redl et al. (Redl et al., 1999). It is interesting how no  
270 significant effect of the sugar (sucrose or trehalose) was observed on the viscoelastic properties,  
271 although slightly superior viscoelastic moduli values could be determined for sucrose-containing  
272 blends.

273 Loss tangent ( $\tan \delta$ ) decreased during the whole temperature test (30 - 120 °C), displaying a  
274 steeper decay at the higher temperature region. This event can be explained by the milder decrease  
275 presented by  $E'$  (about an order of magnitude) when compared to  $E''$  (about two orders of  
276 magnitude), enhancing the solid-like behaviour of the samples as they were heated. As no  
277 distinctive maximum can be easily distinguished in the obtained results, no clear information  
278 regarding the glass transition temperature ( $T_g$ ) of these plastic materials could be withdrawn from  
279 the  $\tan \delta$  profiles. When comparing extruded samples to the rest, general lower  $\tan \delta$  values could  
280 be distinguished, being related to the slightly greater viscoelastic properties already observed.  
281 Regarding the effect of the presence of disaccharides on this profile for samples obtained through  
282 internal mixing, the presence of sugars in samples generally resulted in higher  $\tan \delta$  values  
283 compared to the reference sample at all mixing rates studied. This would be an agreement with the

284 plasticizer role already commented that sugars may possess (Ghanbarzadeh et al., 2006), which  
 285 would promote the viscous character of the blend. However, even if the plasticizer effect of the  
 286 sugars could be implied by the observed lowering of the viscoelastic moduli, no great differences  
 287 in  $\tan \delta$  among the formulations were perceived in extruded samples, being virtually independently  
 288 of the nature of the disaccharide.

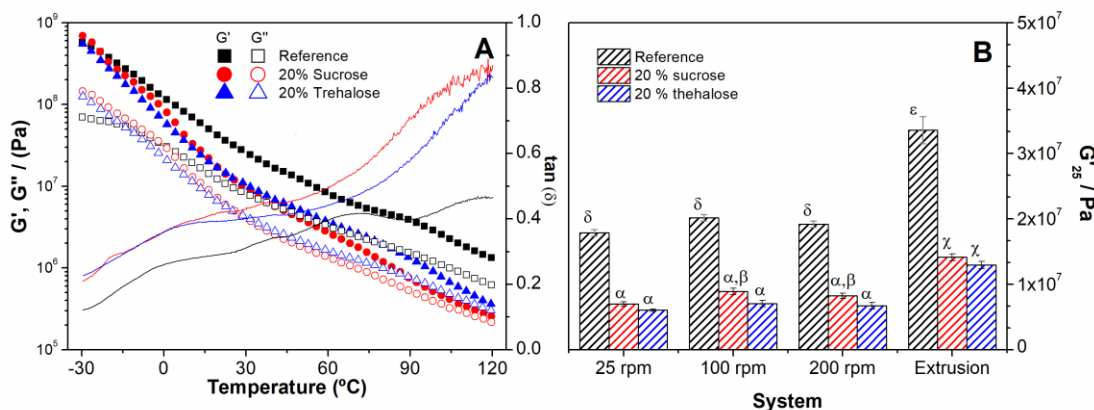


289  
 290 Figure 2. Values from dynamic mechanical analysis (DMA) temperature ramp measurements performed at  
 291 constant frequency (1Hz) and heating rate ( $5\text{ }^{\circ}\text{C min}^{-1}$ ) for the systems: A) 25 rpm for 10 min., (B) 100 rpm for 10  
 292 min., (C) 200 rpm for 5 min. and y (D) extrusion.

293 *Plastics*

294 The viscoelastic properties of the injection moulded gluten-based plastics (reference and sucrose  
295 o trehalose containing samples) were evaluated through temperature sweep tests in torsional mode  
296 from -30 to 120 °C within the LVR. As happened with blends, a continuous decrease was observed  
297 for all samples, not finding any plateau at higher temperatures this time, as may be observed in  
298 Figure 3A for plastics from extruded blends. This denoted the greater thermoplasticity of plastic  
299 samples when compared to the original blends. Lower  $\tan \delta$  values were again detected for the  
300 disaccharide free sample being related to the plasticizing effect of the sugars (Ghanbarzadeh et al.,  
301 2006). In this case, a tendency onto  $\tan \delta$  maximum around 120 °C was suggested, with a smaller  
302 shoulder at lower temperatures.

303 When observing how the elastic moduli,  $G'$ , is affected by the presence of a disaccharide in the  
304 formulation of gluten-based plastics (Figure 3B), an apparent decrease may be distinguished by  
305 either sucrose or trehalose, corroborating the already mentioned plasticizer effect of the  
306 disaccharide. Although no significant differences may be found between sucrose or trehalose, it  
307 seems as lower  $G'$  values were obtained for trehalose systems. Although no significant effect of  
308 the mixing rate could be implied from these results, a slight increase in  $G'$  may be appreciated for  
309 extruded samples, which could be explained by a higher structuration degree when applying a  
310 greater amount of mixing energy.



311  
 312 Figure 3. Linear viscoelastic properties in torsion mode for gluten based injection moulded plastics from  
 313 extruded blends (A); and storage modulus ( $G'$ ) at 1 Hz and 25 °C ( $G'_{25}$ ) for gluten based injection moulded plastics  
 314 obtained from internally mixed blends at 25, 100 and 200 rpm, and from extruded blends (B). Standard deviations  
 315 are presented as error bars (3 technical repeats).

### 316 3.3. Water uptake and effect of the immersion time

317 The water content of all samples was  $13.4 \pm 1.3$  %, which approximately matches the water  
 318 content stated in Table 1. Therefore, no major loss of water has taken place during the sample  
 319 processing, as it is now stated in the revised manuscript.

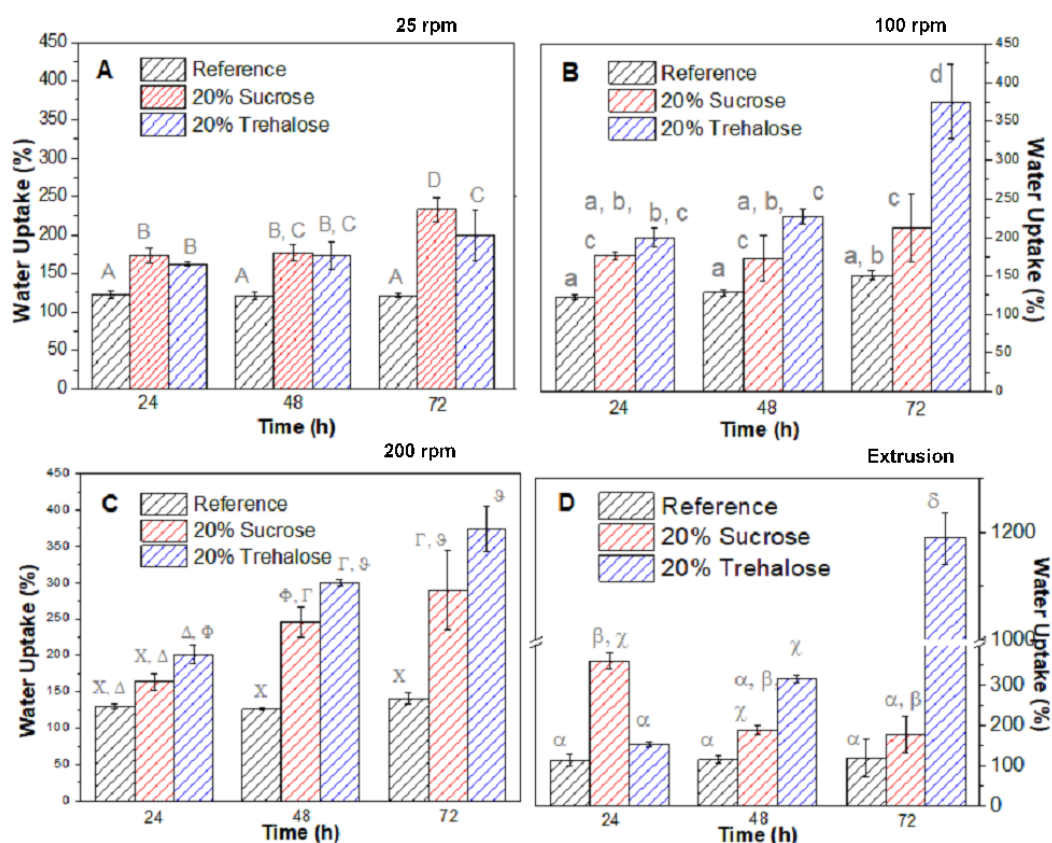
320 The water uptake capacity (WUC) and soluble mater loss (SML) values for plastic materials  
 321 obtained from blends processed though different mixing devices (internal mixing at 25, 100 and  
 322 200 rpm (Figure 4A, 4B and 4C) or extrusion (Figure 4D)) are shown in Figure 4 and Table 2 for  
 323 different immersion times (24, 48 and 72 h).

324 A slight trend towards higher WUC values can be observed when one of the two disaccharides  
 325 was introduced in the composition of the plastics, independently of the immersion time. This  
 326 increase may be related to two different processes happening within the matrix: first, the

327 solubilization of protein onto the aqueous media which would promote an increase in the porosity  
328 in the structure, facilitating the penetration of water (Álvarez-Castillo et al., 2018); and secondly,  
329 the plasticizer effect exerted by the saccharides may diminish the interactions between polymeric  
330 chains, finally reducing the crosslinking during the processing and promoting the swelling of the  
331 matrix. Previous studies have highlighted that moulding gluten samples at temperatures of 130 °C  
332 favoured crosslinking dominated by disulphide bonds (Jansens et al., 2013). Thus, the crosslinking  
333 degree is a parameter that has been closely related to the water uptake capacity of matrices, existing  
334 an equilibrium between the strength of the protein matrix and the swelling resulting from water  
335 molecules penetrating into it (Buonocore et al., 2003). Consequently, a greater crosslinking  
336 hampers the formation of a porous structure, inhibiting the swelling and water absorption of  
337 samples (Álvarez-Castillo et al., 2018). This might be the reason why most of the samples studied  
338 in the present manuscript showed low water uptake capacities. However, it is worth pointing out  
339 that the WUC was noticeably high when the added disaccharide was trehalose. Thus, samples with  
340 20 wt. % of trehalose could even duplicate the WUC value of the reference plastic sample (e.g. 72  
341 h as immersion time for extruded samples, or for samples internal mixed at 100 or 200 rpm). The  
342 fact that no effect of trehalose addition on WUC was observed at lower rates (i.e., 25 rpm) when  
343 using the internal mixer could be related to a poor mixing. On the other hand, the enhancement of  
344 the water absorption by the presence of trehalose at the rest of mixing conditions may be related  
345 to the protecting properties reported to trehalose, which has also been found to be very effective  
346 in the stabilization of proteins during exposure to high temperatures (Kaushik and Bhat, 2003),  
347 which might hinder crosslinking and eventually be the reason behind the higher WUC values  
348 found.

349       When observing the effect of immersion time, WUC values always displayed a remarkable  
350 increase with time when samples contained a disaccharide, while the WUC values remained  
351 unaffected with the immersion time for those samples taken as reference, keeping a nearly constant  
352 value around 115 % in any case. Furthermore, the mixing rate used in internal mixing did directly  
353 influence the absorption capacity of plastics, as higher WUC values were obtained when 200 rpm  
354 was employed. Thus, for those samples obtained from blends mixed at different rates, an increase  
355 of WUC for longer immersion times was generally found, being slighter for lower rates (25 rpm,)   
356 and more apparent for samples mixed at higher rates (100, 200 rpm). Thus, in general terms, it  
357 seems that a lower specific mechanical energy applied through blend mixing promoted a greater  
358 WUC, especially in the presence of disaccharide. On the other hand, the greatest WUC was  
359 achieved for the gluten-based sample obtained by extrusion, which could absorb more than 10  
360 times its original dry weight after 72 h of immersion when trehalose was included. For this system,  
361 a different absorption kinetic was detected depending on the formulation: the sugarless reference  
362 system kept virtually the same WUC value for all immersion times studied, which would imply  
363 that equilibrium was reached before than for the sugar-containing samples; when sucrose was used,  
364 the equilibrium needed longer times to be achieved, as WUC first increased up to 350 % after 24  
365 h of immersion, eventually decreasing to 200 % for longer times; on the other hand, trehalose-  
366 containing systems displayed upwards evolution of WUC with immersion time, eventually  
367 reaching a WUC around 1200 %, which surpasses the threshold defined for superabsorbent  
368 materials (Figure 4D). Other proteins, such as soy or porcine plasma protein, have been used in  
369 the development of superabsorbent materials through different strategies (Álvarez-Castillo et al.,  
370 2020b, 2020a, 2018), mostly at relatively low moulding temperatures (lower than 100 °C). Most  
371 of commercial superabsorbent materials are not renewable, nor sustainable, nor biodegradable, as

372 they are acrylic based. This causes a negative environmental impact which may be avoided if  
 373 natural biodegradable components are used (Álvarez-Castillo et al., 2018). Furthermore, gluten is  
 374 a by-product of the starch industry, which would enhance the sustainability of the process. those  
 375 protein sources are rich in hydrophilic polar residues, such as aspartic or glutamic acid, which are  
 376 not predominant in gluten (Stauffer, 2007). Moreover, gluten-based plastics must be processed at  
 377 higher temperatures, not observing superabsorbent character for gluten plastics as crosslinking is  
 378 promoted. Notwithstanding, the presence of trehalose enhanced water absorption in the conditions  
 379 mentioned.



380

381 Figure 4. Evolution of water absorption capacity (%) after immersion for 24, 48 and 72 h for the systems: (A)  
 382 25 rpm for 10 min., (B) 100 rpm for 10 min., (C) 200 rpm for 5 min. and (D) extrusion. Standard deviations are  
 383 presented as error bars (3 technical repeats).

384  
 385 The values of Soluble matter loss (SML) of the plastic materials (reference, 20 % sucrose and  
 386 20 % trehalose) obtained through different mixing devices and subjected to different immersion  
 387 times (24, 48 and 72 hours) are presented in Table 2. In most cases, SML was not greater than the  
 388 amount of plasticizer originally contained in the formulation, neither considering only glycerol  
 389 and water (33 %), nor when also including the sugars (46.7 %). Only when immersion time was  
 390 72 h for samples containing sugars, a small portion of the protein fraction would be solubilized.  
 391 This fact would point out that the gluten structure formed after immersion moulding was basically  
 392 insoluble, obtaining a gluten matrix, devoid of plasticizers (Zárate-Ramírez et al., 2014b) after  
 393 immersion. In the case of the reference sample, the SML was kept at approximately 22-29 %, while  
 394 for the systems that contained sucrose or trehalose, a small amount of protein was lost onto the  
 395 immersion media, being greater when immersion time was longer. This increase was associated to  
 396 the fact that the immersion media, in addition to extracting glycerol, facilitated the sugar loss.

| Systems       | Immersion Time (h) | Soluble matter loss (%) |            |            |            |
|---------------|--------------------|-------------------------|------------|------------|------------|
|               |                    | 25 rpm                  | 100 rpm    | 200 rpm    | Extrusion  |
| Reference     | 24                 | 22.7 ± 2.6              | 21.4 ± 1.6 | 20.7 ± 0.2 | 21.1 ± 0.4 |
| 20% Sucrose   |                    | 41.7 ± 8.5              | 22.3 ± 0.9 | 43.7 ± 0.2 | 42.8 ± 1.2 |
| 20% Trehalose |                    | 36.8 ± 4.8              | 22.0 ± 0.9 | 40.6 ± 0.5 | 32.5 ± 0.1 |
| Reference     | 48                 | 20.0 ± 0.3              | 44.9 ± 2.7 | 20.5 ± 0.2 | 23.4 ± 0.3 |
| 20% Sucrose   |                    | 42.7 ± 8.0              | 45.6 ± 1.6 | 46.9 ± 1.1 | 40.7 ± 0.6 |
| 20% Trehalose |                    | 44.9 ± 2.0              | 49.4 ± 2.0 | 43.9 ± 2.8 | 40.9 ± 5.9 |

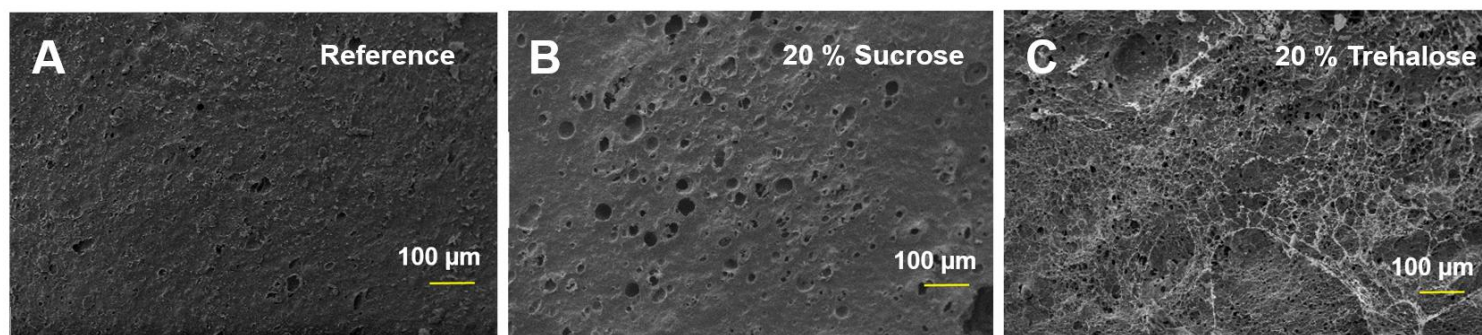
|                      |    |            |            |            |            |
|----------------------|----|------------|------------|------------|------------|
| <b>Reference</b>     | 72 | 22.2 ± 0.5 | 40.5 ± 1.3 | 23.6 ± 4.2 | 24.9 ± 0.3 |
| <b>20% Sucrose</b>   |    | 44.3 ± 4.9 | 42.5 ± 2.5 | 49.8 ± 2.3 | 33.9 ± 5.5 |
| <b>20% Trehalose</b> |    | 48.3 ± 5.7 | 45.8 ± 1.5 | 47.9 ± 3.1 | 52.9 ± 0.6 |

397 Table 2. Soluble matter loss (%) after immersion for 24, 48 and 72 h for the systems: 25 rpm for 10 min., 100  
 398 rpm for 10 min., 200 rpm for 5 min. and extrusion. Standard deviations are presented preceded by ± (3 technical  
 399 repeats).

400

### 401 3.4. Scanning Electron Microscopy

402 This methodology can provide some information about the morphology and microstructure of  
 403 the plasticizer-free matrices resulting from the lyophilisation of the swollen gluten-based plastic  
 404 after immersion. Thus, in Figure 5 the SEM images of the matrices obtained after 72 h immersion  
 405 of gluten-based reference sample (A), and in the presence of a disaccharide in their composition  
 406 (20 wt. % sucrose or trehalose (B and C, respectively)) are shown. It might be observed how both  
 407 the presence of either sucrose or trehalose in the matrix enhanced the formation of a more porous  
 408 structure which could have been promoted by the solubilization of sugars in the immersion media.  
 409 Hence, a porous structure could help with the penetration and holding of water, which is coherent  
 410 with the higher WUC values resulted when sugars were included in the formulation. A greater  
 411 number of pores could be observed in the matrix obtained for the plastic containing 20 wt. % of  
 412 trehalose, with an estimated superficial porosity of approximately 22.5 %, compared to porosities  
 413 of around 3.9 % and 9.4 % presented by reference and containing-sucrose samples, respectively.  
 414 The much greater porosity estimated for the trehalose system would result from the superabsorbent  
 415 character that this sample displayed, as it has been found in previous studies for analogous protein  
 416 superabsorbent materials (Álvarez-Castillo et al., 2018).



417 Figure 5. SEM images (x250) acquired after 72 h of immersion for plastic blended at 200 rpm during 5 minutes: (A)  
418 reference, (B) 20% of sucrose and (C) 20% of trehalose.

#### 419 4. Conclusions

420 The presence of sugars in the formulation of gluten-based plastics did influence the temperature  
421 profile during the mixing stage prior injection moulding. Thus, a greater shear-induced heating of  
422 the sample during processing was generally observed when no sugar was added in the formulation  
423 of gluten-based plastics processed through extrusion. On the other hand, a lag phase was  
424 commonly detected initially in both torque and temperature when using an internal mixer, before  
425 an increase in both parameters took place. The presence of a disaccharide did not significant differ  
426 at a certain rotation speed, as all samples reached nearly the same steady values. Lower torque and  
427 higher temperatures were observed with increased rotation speed, which could be presumably due  
428 to the existence of higher friction forces inside the mixer.

429 When temperature sweep tests were performed on blends, all samples displayed a similar evolution  
430 for the viscoelastic properties, independently of the presence of disaccharide. Thus, a continuous  
431 decrease of both viscoelastic moduli ( $E'$  and  $E''$ ) along with heating was perceived. Lower  
432 viscoelastic moduli values were obtained for disaccharide-containing blends, especially when

433 trehalose was included. Furthermore, higher  $\tan \delta$  values were observed in the presence of a  
434 disaccharide, explained based on sugars plasticizing effect.

435 Gluten-based blends were furtherly processed through injection moulding, displaying quite similar  
436 performances for the viscoelastic moduli, as they continuously decreased as higher temperatures  
437 were reached. Additionally, the plasticizing effect of the disaccharide resulted again in a reduction  
438 of both viscoelastic moduli and a remarkable rise of  $\tan \delta$ . Water uptake capacity of injection  
439 moulded plastics followed three distinct trends: i) the presence of a disaccharide promoted the  
440 absorption capacity of gluten-based plastics, specially the trehalose; ii) longer immersion times  
441 resulted in higher values of water uptake ; iii) the mixing procedure affected the water absorption  
442 of injection moulded plastics, as blends obtained through extrusion gave rise to plastics with higher  
443 water uptakes capacities, even leading to the formation of a superabsorbent material for the sample  
444 with 20 % of trehalose after 72 h of immersion. It should be considered that a small percentage of  
445 protein was solubilized during long immersions (e.g. 72 h). SEM micrographs showed a more  
446 porous and open structure when disaccharides were included in the formulation, which promoted  
447 the formation of voids within the matrix structure, finally enhancing water holding.

448 Thus, the presence of disaccharides in gluten-based plastics exerted a plasticizer role, diminishing  
449 the friction inside the mixer device, and finally giving rise to plastics with lower viscoelastic  
450 moduli and higher  $\tan \delta$ . Thus, swelling was ultimately promoted in the presence of sucrose and,  
451 specially, trehalose, which could have enormous potential when searching for environmentally-  
452 friendly superabsorbent materials.

## 453 **5. Acknowledgements**

454 The authors would like to thank Spanish Ministerio de Ciencia e Innovación – Agencia Estatal de  
455 Investigación (MICINN) and FEDER programme for the financial support provided through the  
456 funding of the RTI2018-097100-B-C21 project. They would also like to thank the full access and  
457 assistance provided to the LECO-CHNS-932 and Zeiss Evo by the Microanalysis and Microscopy  
458 Service (CITIUS-Universidad de Sevilla).

459 **6. References**

- 460 Álvarez-Castillo, E., Bengoechea, C., Guerrero, A., 2020a. Composites from by-products of the food  
 461 industry for the development of superabsorbent biomaterials. *Food Bioprod. Process.* 119, 296–305.  
 462 <https://doi.org/https://doi.org/10.1016/j.fbp.2019.11.009>
- 463 Álvarez-Castillo, E., Bengoechea, C., Guerrero, A., 2020b. Effect of pH on the properties of porcine  
 464 plasma-based superabsorbent materials. *Polym. Test.* 85.  
 465 <https://doi.org/10.1016/j.polymertesting.2020.106453>
- 466 Álvarez-Castillo, E., Bengoechea, C., Rodríguez, N., Guerrero, A., 2019. Development of green  
 467 superabsorbent materials from a by-product of the meat industry. *J. Clean. Prod.* 223, 651–661.  
 468 <https://doi.org/10.1016/j.jclepro.2019.03.055>
- 469 Álvarez-Castillo, E., Del Toro, A., Aguilar, J.M., Guerrero, A., Bengoechea, C., 2018. Optimization of a  
 470 thermal process for the production of superabsorbent materials based on a soy protein isolate. *Ind.*  
 471 *Crops Prod.* 125, 573–581. <https://doi.org/10.1016/j.indcrop.2018.09.051>
- 472 Buonocore, G.G., Del Nobile, M.A., Panizza, A., Corbo, M.R., Nicolais, L., 2003. A general approach to  
 473 describe the antimicrobial agent release from highly swellable films intended for food packaging  
 474 applications. *J. Control. Release* 90, 97–107. [https://doi.org/https://doi.org/10.1016/S0168-](https://doi.org/https://doi.org/10.1016/S0168-3659(03)00154-8)  
 475 [3659\(03\)00154-8](https://doi.org/https://doi.org/10.1016/S0168-3659(03)00154-8)
- 476 Cho, S.-W., Gällstedt, M., Johansson, E., Hedenqvist, M.S., 2011. Injection-molded nanocomposites and  
 477 materials based on wheat gluten. *Int. J. Biol. Macromol.*
- 478 Drobny, J.G., 2014. 4 - Processing Methods Applicable to Thermoplastic Elastomers, in: Drobny,  
 479 J.G.B.T.-H. of T.E. (Second E. (Ed.), *Plastics Design Library*. William Andrew Publishing, Oxford,  
 480 pp. 33–173. <https://doi.org/https://doi.org/10.1016/B978-0-323-22136-8.00004-1>
- 481 Emin, M.A., Schuchmann, H.P., 2013. Analysis of the dispersive mixing efficiency in a twin-screw  
 482 extrusion processing of starch based matrix. *J. Food Eng.* 115, 132–143.  
 483 <https://doi.org/https://doi.org/10.1016/j.jfoodeng.2012.10.008>
- 484 Fang, Y., Zhang, B., Wei, Y., 2014. Effects of the specific mechanical energy on the physicochemical  
 485 properties of texturized soy protein during high-moisture extrusion cooking. *J Food Eng* 121, 8–32.
- 486 Felix, M., Romero, A., Cordobes, F., Guerrero, A., 2015. Development of crayfish bio-based plastic  
 487 materials processed by small-scale injection moulding. *J. Sci. Food Agric.* 95, 679–687.  
 488 <https://doi.org/10.1002/jsfa.6747>
- 489 Food and Agricultural Organization of the United Nations, 2019. Biannual Report of Global Food  
 490 Markets.
- 491 Ghanbarzadeh, B., Oromiehie, A.R., Musavi, M., D-Jomeh, Z.E., Rad, E.R., Milani, J., 2006. Effect of  
 492 plasticizing sugars on rheological and thermal properties of zein resins and mechanical properties of  
 493 zein films. *Food Res. Int.* 39, 882–890. <https://doi.org/10.1016/j.foodres.2006.05.011>
- 494 Gómez-Heincke, D., Martínez, I., Partal, P., Guerrero, A., Gallegos, C., 2016. Development of  
 495 antimicrobial active packaging materials based on gluten proteins. *J. Sci. Food Agric.*  
 496 <https://doi.org/10.1002/jsfa.7525>
- 497 Gómez-Martínez, D., Barneto, A.G., Martínez, I., Partal, P., 2011. Modelling of pyrolysis and combustion  
 498 of gluten–glycerol-based bioplastics. *Bioresour. Technol.* 102, 6246–6253.  
 499 <https://doi.org/10.1016/j.biortech.2011.02.074>
- 500 Hernandez-Izquierdo, V.M., Krochta, J.M., 2008. Thermoplastic Processing of Proteins for Film  
 501 Formation—A Review. *J. Food Sci.* 73, R30–R39. [https://doi.org/10.1111/j.1750-](https://doi.org/10.1111/j.1750-3841.2007.00636.x)  
 502 [3841.2007.00636.x](https://doi.org/10.1111/j.1750-3841.2007.00636.x)
- 503 Hosney, R.C., Rogers, D.E., 1990. The formation and properties of wheat flour doughs. *Crit. Rev. Food*  
 504 *Sci. Nutr.* 29, 73–93. <https://doi.org/10.1080/10408399009527517>
- 505 Jansens, K.J.A., Vo Hong, N., Telen, L., Brijs, K., Lagrain, B., Van Vuure, A.W., Van Acker, K.,  
 506 Verpoest, I., Van Puyvelde, P., Goderis, B., Smet, M., Delcour, J.A., 2013. Effect of molding

- 507 conditions and moisture content on the mechanical properties of compression molded glassy, wheat  
508 gluten bioplastics. *Ind. Crops Prod.* 44, 480–487. <https://doi.org/10.1016/j.indcrop.2012.10.006>
- 509 Jerez, A., Partal, P., Martínez, I., Gallegos, C., Guerrero, A., 2005. Rheology and processing of gluten  
510 based bioplastics. *Biochem. Eng. J.* 26, 131–138. <https://doi.org/10.1016/j.bej.2005.04.010>
- 511 Jiménez-Rosado, M., Zarate-Ramírez, L.S., Romero, A., Bengoechea, C., Partal, P., Guerrero, A., 2019.  
512 Bioplastics based on wheat gluten processed by extrusion. *J. Clean. Prod.*
- 513 Kajiwara, T., Nakayama, Y., 2011. Capturing the Efficiency of a Melt-Mixing Process for Polymer  
514 Processing. *J. Chem. Eng. JAPAN* 44, 831–839. <https://doi.org/10.1252/jcej.11we081>
- 515 Kaushik, J.K., Bhat, R., 2003. Why Is Trehalose an Exceptional Protein Stabilizer? *J. Biol. Chem.* 278,  
516 26458–26465. <https://doi.org/10.1074/jbc.M300815200>
- 517 Redl, Andréas, Morel, M.H., Bonicel, J., Guilbert, S., Vergnes, B., 1999. Rheological properties of gluten  
518 plasticized with glycerol: Dependence on temperature, glycerol content and mixing conditions.  
519 *Rheol. Acta.* <https://doi.org/10.1007/s003970050183>
- 520 Redl, Andreas, Morel, M.H., Bonicel, J., Vergnes, B., Guilbert, S., 1999. Extrusion of wheat gluten  
521 plasticized with glycerol: Influence of process conditions on flow behavior, rheological properties,  
522 and molecular size distribution. *Cereal Chem.* 76, 361–370.  
523 <https://doi.org/10.1094/CCHEM.1999.76.3.361>
- 524 Sothornvit, R., Krochta, J.M., 2005. Plasticizers in edible films and coatings, in: *Innovations in Food*  
525 *Packaging.* Elsevier, pp. 403–433. <https://doi.org/10.1016/B978-012311632-1/50055-3>
- 526 Stauffer, C.E., 2007. Principles of Dough Formation BT - Technology of Breadmaking, in: Cauvain,  
527 S.P., Young, L.S. (Eds.), . Springer US, Boston, MA, pp. 299–332. [https://doi.org/10.1007/0-387-](https://doi.org/10.1007/0-387-38565-7_11)  
528 [38565-7\\_11](https://doi.org/10.1007/0-387-38565-7_11)
- 529 Ullsten, N.H., Cho, S.-W., Spencer, G., Gällstedt, M., Johansson, E., Hedenqvist, M.S., 2009. Properties  
530 of Extruded Vital Wheat Gluten Sheets with Sodium Hydroxide and Salicylic Acid.  
531 *Biomacromolecules* 10, 479–488. <https://doi.org/10.1021/bm800691h>
- 532 Zárate-Ramírez, L.S., Martínez, I., Romero, A., Partal, P., Guerrero, A., 2011. Wheat gluten-based  
533 materials plasticised with glycerol and water by thermoplastic mixing and thermomoulding. *J. Sci.*  
534 *Food Agric.* 91, 625–633. <https://doi.org/10.1002/jsfa.4224>
- 535 Zárate-Ramírez, L.S., Romero, A., Bengoechea, C., Partal, P., Guerrero, A., 2014a. Thermo-mechanical  
536 and hydrophilic properties of polysaccharide/gluten-based bioplastics. *Carbohydr. Polym.* 112, 24–  
537 31. <https://doi.org/10.1016/j.carbpol.2014.05.055>
- 538 Zárate-Ramírez, L.S., Romero, A., Martínez, I., Bengoechea, C., Partal, P., Guerrero, A., 2014b. Effect of  
539 aldehydes on thermomechanical properties of gluten-based bioplastics. *Food Bioprod. Process.* 92,  
540 20–29. <https://doi.org/10.1016/j.fbp.2013.07.007>
- 541